Drying of *Citrus grandis* (Pomelo) Fruit Juice Using Block Freeze Concentration and Spray Drying

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Pomelo fruit juice is concentrated using two different techniques, i.e. block freeze concentration (BFC) and thermal concentration (TC), and both concentrates were spray dried separately to observe the effects of two different concentration techniques on the physicochemical properties of the spray dried powder. BFC juice had comparatively better values of retention of acidity, colour, vitamin C, DPPH● scavenging activity, total phenolic content (TPC), and total sugar content than TC juice. The powder obtained from the combination of BFC and spray drying had better retention of vitamin C (121.17 mg/100 g), colour, and exhibited higher DPPH● scavenging activity than TC spray dried powder. The physical parameters including solubility index, bulk density, and hygroscopicity were significantly higher in BFC spray dried powder. BFC was observed as an efficient concentration technique in comparison to TC as a pretreatment to produce spray dried powder with a better retention of bioactive components and powder flow properties.

**Keywords:** pomelo juice powder, block freeze concentration, thermal concentration, spray drying

Pomelo (*Citrus grandis* Linn.), a monoembryonic species of the genus *Citrus*, contains a plethora with nutrients (Bharali & Saikia, 2004). Beside nutraceutical ingredients such as carotenoids, lycopene, anthocyanin, limonoids, naringin, and water-soluble vitamins, the fruit is also rich in crude fibre and dietary fibre (Bohra & Srinivas, 2015). Being seasonal in nature, availability of the fruit is limited up to 3–4 months in a year. Production of fruit juice powder is a dehydration technique, which eases storage, transportation, and increases the shelf life up to several months.

Spray drying produces regular and spherical shaped powder particles through the atomisation of fine liquid droplets (Fazaeli et al., 2012). Spray drying of pomelo fruit juice may be a good alternative to make it available throughout the season. The feedstock used for spray drying is usually a pre-concentrate up to 25–30% for lab-scale spray dryer. Application of heat for pre-concentration adversely affects the physicochemical properties of the juice, so a comparatively inferior quality fruit powder is produced (Murugesan & Orsat, 2011). Thermal concentration of juice is generally performed by the application of heat energy in batches or in a continuous process (Adnan et al., 2018). Block freeze concentration (BFC) is a technique where free water can be removed from the feed solution in the form of ice (Pitzold et al., 2015). In this technique, the liquid sample is frozen until the temperature at the centre of the product reaches below the freezing point of the sample, and then the free water from the sample is removed through partial thawing and centrifugation.

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The present study is undertaken to elucidate the effect of two different pre-concentration techniques (BFC and thermal concentration (TC)) on the physicochemical and antioxidant properties of spray dried pomelo juice powder. As per the best of our knowledge, we are reporting the effect of combination of two different concentration techniques and spray drying on the physicochemical and phytochemical properties of the powder the first time.

1. Materials and methods

1.1. Sample preparation

Pomelo fruit (collected from campus of Tezpur University, India) juice was extracted from peeled fruit using lab scale juicer (Philips HR 1863/20, India) and was filtered by 3-layered cheesecloth. The extracted juice was stored at 5 °C for further analysis.

1.2. Chemical analysis of pomelo juice

Moisture content, titratable acidity, pH, total soluble solids (TSS), and vitamin C content were calculated as per the method of AOAC (1990) official methods such as AOAC 930.15; AOAC, 942.15; AOAC 981.12; AOAC, 932.14, and AOAC 967.21, respectively. Naringin and total phenolic content was determined according to the method given by Davis (1947) and Mishra and co-workers (2014), respectively.

1.3. Overview of experimental method

Figure 1 illustrates an overall experimental procedure, where pomelo juice was frozen in a deep freezer (Lyolab_3S, Lyophilisation Systems India Pvt Ltd.) at –20 °C and then centrifuged (Eppendorf, Hamburg) to separate the solutes from the solid ice. The fraction received after the first block freeze concentration (BFC) cycle was used as feed solution for the second BFC cycle, and similarly, the second concentrated fraction was used for the third cycle of concentration (Petzold et al., 2015).

1.4. Freezing time calculation and centrifugation of pomelo juice

Juice (25 ml) was frozen in static freezer at –20 °C (temperature during freezing was calculated using needle type digital thermocouple) for 12 h, and the freezing time of juice was estimated using Planck’s equation.

\[
t_f = \frac{L \rho a}{T_f - T_a \left(\frac{P}{h} + \frac{R}{k} \right)}
\]

where,

- \(t_f\) = freezing time (h)
- \(L\) = latent heat of fusion of the food (L =333.2 kJ kg\(^{-1}\))
- \(\rho\) = density of the juice (density of juice 1232 kg m\(^{-3}\) at 25 °C)
- \(P\) = constant (for cylinder =\(1/4\))
- \(R\) = constant (for cylinder =\(1/16\))
- \(a\) = thickness of the material (m)
- \(T_f\) = freezing temperature (°C) (calculated by needle-based thermocouple)
- \(T_a\) = freezing temperature of medium (°C)

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h = convective heat transfer coefficient at the surface of the food \( (h = 7.26 \text{ W m}^{-2}\text{ k}^{-1}) \)

\( k_f = \) thermal conductivity of the frozen food \( (k = 1.6 \text{ W m}^{-2}\text{ k}^{-1}) \)

\( L, h, \) and \( K_f \) value were calculated as per the method of Ikegwu and Ekwu (2009). Separation of solutes from the frozen juice was carried out by centrifugation at 5400 rotation centrifugal force (rcf) for 25 min at 20 °C. After centrifugation, the concentrated juice was collected, and the remaining frozen core was melted to estimate the soluble solids of both fractions.

![Fig. 1. Generalised scheme of overall experimental procedure](image)

1.5. Analysis of juice concentrate

1.5.1. Percentage of concentrate. Percentage of concentrate (\% PC) was estimated for each BFC using the following equation:

\[
\%PC = \frac{W_i - W_f}{W_i}
\]

(2) (Petzold et al., 2015)

where, \( W_i \) is initial weight of the frozen fraction and \( W_f \) is the final weight of the frozen fraction.
1.5.2. Concentration efficiency. Equation 3 was used to calculate the concentration efficiency of the block freeze concentration:

\[
\eta \% = \frac{C_s - C_f}{C_s} \times 100 \tag{3} \ (\text{PETZOLD et al., 2015})
\]

where, \( C_s \) and \( C_f \) are the concentrations of soluble solids (°Brix) in the juice and frozen fraction, respectively.

1.5.3. Recovery of solute. Recovery of solutes was quantified using the equation 4:

\[
Y = \frac{C_s \times M_s}{C_o \times M_o} \tag{4} \ (\text{PETZOLD et al., 2015})
\]

where, \( Y \) = recovery of solutes (kg solute/kg initial solute), \( C_s \) is the soluble solids (°Brix) present in the initial juice of each BFC cycle, \( M_s \) and \( M_o \) are mass (kg) of the concentrated juice and initial juice of each cycle respectively.

1.6. Concentration of juice using thermal concentration (TC)

Freshly extracted pomelo juice was concentrated in a rotary vacuum evaporator (Equitron, Medica Instruments, India) at a speed of 60 rcf and at 60 °C till the soluble solids of the juice concentrate became equivalent to the 3rd cycle BFC juice.

1.7. Spray drying of concentrated pomelo fruit juice

The BFC and TC pomelo juice (250 ml) was spray dried in lab scale spray dryer (SD-Basic LabPlant, England) using maltodextrin (15% w/v of the initial volume of juice). Inlet temperatures and recorded outlet temperatures were 160 °C and 90 °C, respectively. During spray drying compressor pressure, airflow rate, and feed rate were maintained at 0.12 MPa, 75±1.5 m³ h⁻¹, and 13–15 ml min⁻¹, respectively. All formulations for spray drying were performed in triplicates.

1.8. Hygroscopicity, water solubility index (WSI), bulk density (BD), DPPH free radical (DPPH\( \cdot \)) scavenging activity, and colour properties

Hygroscopicity, WSI, BD, and DPPH\( \cdot \) scavenging activity of powders was evaluated as per the method of CAI and CORKE (2000), SUBRAMANIAN and co-workers (1994), and MISHRA and co-workers (2014), respectively. Colour of the spray dried powder was measured using Hunter colour lab (Ultra Scan VIS, USA).

1.9. Statistical analysis

All experiments were performed in triplicates, the obtained data were statistically analysed using IBM SPSS Statistics Version 20.0, Armonk, NY: IBM Corporation package, and the data are presented as the mean ± standard deviation. The means were separated using the Duncan’s multiple range test (P<0.05).
2. Results and discussion

2.1. Effect of centrifugation time for maximum recovery of solutes

Juice used for the present investigation had pH 3.64, 8.6 °Brix, 0.98% acidity, 90 mg/100 ml vitamin C, and 90.37% moisture content. The time for maximum possible recovery of solute during the first cycle of BFC was considered for second and third cycles of BFC (Table 1). The efficiency of concentration progressively increased to 86.95% in 25 min of centrifugation and thereafter decreased due to loss of solid content. Therefore, 25 min for centrifugation was considered as the appropriate time for the maximum possible recovery of solute (0.61 kg of solute/kg of initial solute).

<table>
<thead>
<tr>
<th>Centrifugation time (min)</th>
<th>Efficiency (%)</th>
<th>Recovery of solute (kg kg⁻¹)</th>
</tr>
</thead>
<tbody>
<tr>
<td>5</td>
<td>67.33±0.01f</td>
<td>0.29±0.03f</td>
</tr>
<tr>
<td>10</td>
<td>73.67±0.01e</td>
<td>0.33±0.01e</td>
</tr>
<tr>
<td>15</td>
<td>78.11±0.02d</td>
<td>0.38±0.01d</td>
</tr>
<tr>
<td>20</td>
<td>82.05±0.01c</td>
<td>0.48±0.02c</td>
</tr>
<tr>
<td>25</td>
<td>86.95±0.02a</td>
<td>0.61±0.01b</td>
</tr>
<tr>
<td>30</td>
<td>83.27±0.03b</td>
<td>0.65±0.01a</td>
</tr>
</tbody>
</table>

Values are means ± standard deviation of three determinations (n=3). Values followed by different superscript letter in a column are significantly different (P≤0.05)

2.2. Calculation of recovery of solute, efficiency, and percentage of concentration

Figure 2 presents the recovery of the solute after the first BFC cycle, which was 0.61 kg kg⁻¹ of the solute of the initial solution, and at the end of third BFC cycle, the recovery of solute reached 0.77 kg kg⁻¹ the solute of the initial solution. Due to solid loss in ice, efficiency was reduced from 86.95% to 83.27%. The % concentration of juice increased gradually from 22.90 to 39.32% with the increasing number of cycles of BFC.

2.3. Effect of concentration on the freezing time of pomelo fruit juice

The effect of concentration of pomelo juice on freezing time was calculated using Plank’s equation. A significant difference between the freezing plateaus of fresh and concentrated juices was clearly noticed (Fig. 3). Raw juice could be frozen in 4.61 h, however, the first cycle, second cycle, and third cycle concentrated juices took 4.90, 5.23, and 5.60 h, respectively.
Fig. 2. Percentage efficiency of concentration of solute vs centrifugation time and recovery of solute vs centrifugation time after first BFC cycle
- ▲: Recovery of solutes; ■: efficiency, %

Fig. 3. Freezing curves for raw sample and each BFC cycle
2.4. Effect of BFC and TC on chemical and phytochemical properties of pomelo fruit juice

Moisture content was reduced from 90.37% to 75.12% after the third cycle of BFC, while 74.39% moisture content was recorded in TC juice (Table 2). The TSS increased with the number of cycles and reached 14.7, 17.6, and 22.0% after first, second, and third BFC cycle, respectively. The final °Brix of TC juice was maintained at 22.18 °Brix of concentrated juice. Acidity of BFC juice increased from 0.98 to 1.92% and acidity of TC juice increased only to 1.27%. An increase in acidity in BFC juice may be due to the concentration-effect only, but on the other hand, heating in TC probably affects the composition of acids and hydrogen ion concentration, so increase in acidity was comparatively lower than in BFC juice.

Being unstable in nature, vitamin C gets easily destroyed when exposed to heat during concentration (Sélé et al., 2015), and TC juice had 36.15 mg/100 g of vitamin C only, while freezing showed no adverse effect on vitamin C and a good amount of vitamin C was retained even after the third cycle of BFC (Table 2). The total sugar content of juice increased during BFC (16.33 to 39.32%) and TC (16.33 to 38.12 mg ml⁻¹), no significant difference was observed in total sugar content of concentrates produced by either the technique.

2.5. Effect of BFC and TC on the antioxidant property of juice

Total phenolic content (TPC) of juice increased from 74.36 to 103.10 GAE/100 ml during 1st to 3rd cycles of BFC, TC juice had 101.63 GAE/100 ml of juice for the same degree of concentration (Table 2). Although no significant difference was observed in TPC, BFC retained the DPPH● scavenging activity of the juice increased gradually from 1st to 3rd cycle, however, a reverse trend was observed when concentrated thermally (Table 2). The low temperature used for concentration had not changed the functional properties of phenolic components present in juice and so protected the DPPH● scavenging activity (Mishra et al., 2014), while exposure of higher temperature for concentration unfavourably affected the structure of phenolic content and so reduced the DPPH● scavenging activity.

2.6. Effect of BFC and TC on bitterness of juice

The retention of naringin content was comparatively higher in juice concentrated by heating (5.82 μg ml⁻¹) (Table 2). Being water-soluble in nature, the fraction of naringin was removed from the juice along with the removal of ice, while in the case of TC, naringin could not be eliminated from juice, and due to the concentration effect, the level of naringin increased in the juice. Naringin is a thermally stable component that may resist up to 100–120 °C and above 120 °C it follows first-order reaction (Ioannou et al., 2018).
Table 2. Chemical and phytochemical properties of fresh and concentrated juices

<table>
<thead>
<tr>
<th>Sample</th>
<th>pH</th>
<th>TSS (°Brix)</th>
<th>Acidity (% citric acid)</th>
<th>Moisture content (%)</th>
<th>Vitamin C (mg/100 g)</th>
<th>Total sugar (%)</th>
<th>Naringin content (µg ml⁻¹)</th>
<th>DPPH (% inhibition)</th>
<th>TPC (mg GAE/100 ml)</th>
</tr>
</thead>
<tbody>
<tr>
<td>Raw juice</td>
<td>3.64±0.01ᵇ</td>
<td>8.60±0.00ᵈ</td>
<td>0.98±0.00ᵈ</td>
<td>90.37±0.03ᵇ</td>
<td>90.3±3.29ᵈ</td>
<td>16.3±1.09ᵈ</td>
<td>3.2±0.21ᵃ</td>
<td>79.0±1.00ᵈ</td>
<td>56.69±0.05ᵈ</td>
</tr>
<tr>
<td>1ˢᵗ BFC juice</td>
<td>3.61±0.02ᶜ</td>
<td>14.70±0.00ᶜ</td>
<td>1.04±0.01ᶜ</td>
<td>82.7±0.38ᵇ</td>
<td>95.6±4.54ᶜ</td>
<td>22.9±0.2₀ᶜ</td>
<td>2.8±0.20ᵇ</td>
<td>92.4±0.50ᶜ</td>
<td>74.36±0.55ᶜ</td>
</tr>
<tr>
<td>2ⁿᵈ BFC juice</td>
<td>3.56±0.01ᵈ</td>
<td>17.60±0.00ᵇ</td>
<td>1.28±0.00ᵇ</td>
<td>79.1±0.38ᶜ</td>
<td>105.2±3.15ᵇ</td>
<td>29.0±0.89ᵇ</td>
<td>2.4±0.09ᶜ</td>
<td>95.3±0.57ᵇ</td>
<td>88.04±0.54ᵇ</td>
</tr>
<tr>
<td>3ʳᵈ BFC juice</td>
<td>3.44±0.01ᵉ</td>
<td>22.00±0.00ᵃ</td>
<td>1.92±0.01ᵃ</td>
<td>75.1±0.26ᵈ</td>
<td>39.3±0.33ᵃ</td>
<td>1.8±0.11ᵉ</td>
<td>97.3±0.57ᵃ</td>
<td>103.1±0.56ᵃ</td>
<td>101.63±0.29ᵃ</td>
</tr>
<tr>
<td>TC juice</td>
<td>3.84±0.03ᵃ</td>
<td>22.18±0.32ᵃ</td>
<td>1.27±0.00ᵇ</td>
<td>74.3±0.17ᵃ</td>
<td>36.1±1.59ᵇ</td>
<td>3.8±0.02ᵈ</td>
<td>76.3±0.57ᵃ</td>
<td>101.63±0.29ᵃ</td>
<td></td>
</tr>
</tbody>
</table>

Values are means ± standard deviation of three determinations (n= 3). Values followed by different superscript letter in a column are significantly different (P≤0.05).
2.7. Colour analysis

The unstable nature of anthocyanin is a main concern during the processing and treatment of pomelo juice (Fig. 4). L, hue, and a/b values of raw juice used for present investigation were 15.7, 5.7, and 9.3, respectively, and BFC had values 10.54, 3.72, and 15.47, respectively. Retention of anthocyanin content during BFC retains the redness (high a/b value) of the juice. Anthocyanins being heat-sensitive are destroyed easily when heated and so significant increases in L (48.42) and hue angle (24.78) were noticed. Destruction of anthocyanins and polymerisation of anthocyanins with other polyphenols are the key factors for the reduction of redness of juice during processing.

![Figure 4. Colour parameters of raw juice, BFC juice, TC juice, BFC spray dried powder, and TC spray dried powder](image)

2.8. Comparative study of the effect of two different concentration techniques on physicochemical properties of spray dried powder

Final moisture content of powder produced using a combination of freeze concentration and spray drying (6.06%) was comparatively lower than of TC spray dried powder (6.80%) (Table 3). Low temperature concentration protected the capillaries against damage and so facilitated the faster removal of moisture during spray drying. Damage to structure of capillaries during TC may be one of the feasible reasons of the reduced rate of removal of moisture in TC spray dried powder.
### Table 3. Comparative study of the effect of BFC and TC as pretreatments on the physical and phytochemical properties of spray dried powder

<table>
<thead>
<tr>
<th>Sample</th>
<th>Moisture content (%)</th>
<th>Solubility index (%)</th>
<th>Bulk density (g cm⁻³)</th>
<th>Hygroscopicity (g/100 g)</th>
<th>Vitamin C (mg/100 g)</th>
<th>Naringin content (μg g⁻¹)</th>
<th>DPPH (% inhibition)</th>
<th>TPC (mg GAE 100 g⁻¹)</th>
</tr>
</thead>
<tbody>
<tr>
<td>BFC spray dried powder</td>
<td>6.06±0.58ᵇ</td>
<td>85.50±0.50ᵃ</td>
<td>0.56±0.00ᵇ</td>
<td>18.20±0.10ᵃ</td>
<td>121.17±5.09ᵃ</td>
<td>1.79±0.09ᵇ</td>
<td>95.66±0.57ᵃ</td>
<td>97.56±0.50ᵃ</td>
</tr>
<tr>
<td>TC spray dried powder</td>
<td>6.80±0.20ᵃ</td>
<td>83.37±0.33ᵇ</td>
<td>0.69±0.01ᵇ</td>
<td>15.75±0.04ᵇ</td>
<td>15±3.03ᵇ</td>
<td>2.95±0.02ᵇ</td>
<td>75.66±0.57ᵇ</td>
<td>91.89±0.31ᵇ</td>
</tr>
</tbody>
</table>

Values are means ± standard deviation of three determinations (n=3). Values followed by different superscript letter in a column are significantly different (P≤0.05).
WSI of BFC powder and TC powder were 85.50% and 83.37%, respectively. The powder produced from the concentrates showed good solubility, which indicated its application possibilities in new food product development. BD values of freeze- and the thermally concentrated powders were 0.56 and 0.69, respectively (Table 3). Formation of ice crystals increases porosity of the cell wall, so by drying a porous and fluffy powder is obtained, however, denser powder particles were observed in the case of TC spray dried powder. Denser particles can easily settle into the interstitial spaces resulting high BD (Mishra et al., 2014). The increased porosity of the cell wall also increases the tendency of BFC spray dried powder to absorb moisture during storage (Table 3). Similar findings were observed by Goula and Adamopoulos (2005) during spray drying of tomato pulp.

Vitamin C content of powder produced using a combination of BFC and spray drying was 121.17 mg/100 g (Table 3), while of TC spray dried was 15 mg/100 g only. Being unstable in nature, vitamin C degraded very easily during concentration by heat.

Naringin content of BFC spray dried powder was 1.79 μg ml⁻¹, whereas of TC spray dried powder was 2.95 μg g⁻¹ (Table 3). DPPH• scavenging activity of BFC spray dried powder was significantly higher (95.66%) than that of TC spray dried powder (Table 3). Decreased activity of DPPH• scavenging activity may be due to the breaking down of phenolic structure or synthesis of new different forms of phenolics with poor DPPH• scavenging activity (Mishra et al., 2014). TPC values of BFC spray dried powder and TC spray dried powder were 97.56 mg GAE/100 g and 91.89 mg GAE/100 g, respectively. Reduction in phenolic content may be attributed to high inlet temperature, which causes de-polymerisation and degradation of polyphenolic compounds.

Loss of redness in TC spray dried powder was due to the degradation and rapid oxidation of anthocyanins during concentration. BFC protected the anthocyanin content from oxidation and so BFC spray dried powder showed comparatively better retention of anthocyanins, which correlates with the lower hue angle and comparatively high a/b value (Fig. 4).

3. Conclusions

Concentration by BFC retained the phytochemical and colour properties of the pomelo fruit juice. Tuning to TC decreased colour values and physicochemical attributes. The powders produced by the two different pre-treatment techniques were significantly different on the basis of functional and physicochemical properties. Combination of BFC and spray drying provided a good quality powder with potent antioxidant properties, retention of anthocyanins, and better solubility index than of TC spray dried powder. Naringin content was significantly reduced in BFC concentrate and BFC spray dried powder, gaining increased acceptability among consumers. In conclusion, BFC as a pre-treatment permits the production of juice concentrate with higher retention of bioactive nutrients proving it a promising technique in the preservation of heat-labile constituents of juice.

References


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