1 2 3	This is the final version (unedited) of the paper published in <u>Journal of Chromatography A</u> <u>Volume 1232</u> , 6 April 2012, Pages 3–18 doi:10.1016/j.chroma.2011.11.049
45	http://www.sciencedirect.com/science/article/pii/S002196731101750X
6	
7	Overpressured layer chromatography: from the pressurized ultramicro chamber to
8	BioArena system
9	
10	Ernő Tyihák ¹ , Emil Mincsovics ² and Ágnes M. Móricz ¹
11	
12	1) Plant Protection Institute, Hungarian Academy of Sciences, Budapest, Herman O. Str. 15,
13	POB 102, H-1525 Hungary
14	2) OPLC-NIT Co. Ltd, Budapest, Hungary and Corvinus University, Faculty of
15	Horticultural Sciences, Department of Genetics and Plant Breeding, Budapest, Hungary
16	
17	Corresponding author: Ernő Tyihák, Tel.: 0036-1-4877-515; Fax: 0036-1-4877-555; e-mail:
18	etyih@nki.hu
19	
20	Abstract
21	The pressurized ultramicro (UM) chamber as a closed adsorbent layer chamber enables the use
22	of a special chromatoplate and a pump to increase and optimize the mobile phase flow velocity
23	through an optional development distance in an adsorbent layer. This chamber is the basic
24	instrument of overpressured-layer chromatography (OPLC), which is a separation technique that
25	combines the advantages of conventional TLC/HPTLC with those of HPLC. The versions of
26	OPLC instrument, the character and achievement of off-line and on-line OPLC systems in
27	analytical and preparative use are described.

The development of BioArena as a complex bioautographic system means an exploitation of the unique advantages of planar-layer system for detection, isolation and identification of new antimicrobials, antineoplastics, biopesticides and other biologically active substances as well as for studying fundamental biochemical reactions and mechanisms.

32

33 1. Introductory remarks

34

Different directions of chromatography from its discovery have undergone innovative developments. Among the technical versions one of the most important is liquid chromatography (LC), which nowadays includes two basic techniques – column LC (CLC) and layer LC (LLC).
CLC and LLC techniques – as supplementary techniques due to their similar theoretical basis and technical arrangements – have from the beginnings been progressed similarly.

40 It is known that layer liquid chromatographic techniques are rather simple and require in general 41 few appliances. Paper chromatography (PC) and its various versions, developed in the 1940s by 42 Martin and his colleagues [1], have to be mentioned first. The technical solutions of PC laid the 43 foundation of modern amino acid and peptide analysis and promoted the discovery of the structure of insulin [1-3]. However, the rapid progress in different fields after 2nd world war 44 45 demanded the elimination of the limitations of PC (e.g. limited adsorption of paper [4]). Among 46 the innovative developments the discovery of thin-layer chromatography (TLC) by Ismailov 47 and Shraiber [5] as well as Békésy [6], the improvement by Kirchner et al. [7], additionally the 48 standardization and spreading by Stahl et al. [8], opened new horizons in separation science and 49 contributed to the isolation and analysis of many natural and synthetic substances. Today, 50 versions of this classical layer liquid chromatographic technique are indispensable in various 51 fields of scientific research and practice, but the limitations of TLC (e.g. limited efficiency) 52 demanded further development efforts.

53 It is logic to use an adsorbent layer of fine particle permitting effective separation similar to 54 HPLC technique [9]. High-performance thin-layer chromatography (HPTLC) is based already on 55 the use of chromatoplates coated with fine-particle-size and narrow particle-size distribution 56 adsorbent and instrumentation [10, 11]. Comparison of TLC and HPTLC chromatoplates 57 revealed that the latter permitted better separation, higher efficiency, but in a short development 58 distance only, because the migration of the mobile phase was slowed down and therefore, the 59 diffusion of the separated components on the adsorbent layer was increased using a relatively 60 long solvent migration distance, especially if viscous mobile phases were applied. The quadratic 61 development law of classical TLC is also valid in HPTLC [12].

HPTLC is in certain cases considered as the planar layer version of HPLC [13, 14]. This is true in part only, since a very important and indispensable step of HPLC is missing in the application of fine particle-size adsorbent layer with a narrow particle-size distribution. Namely, the eluent is dispensed into the system by forced flow (by pressure) in the case of HPLC [9]. That is HPTLC is a half solution to the efficient modern planar layer liquid chromatography.

67 The steady stream of innovations in layer liquid chromatography has resulted in most diverse 68 technical versions in the last decades. In these modern technical versions the mobile phase 69 migrates through the stationary phase in the layer arrangement under the additional action of 70 forced flow. This can be achieved especially successfully and attractively by application of a 71 pressurized ultramicro (UM) chamber and a pump system in overpressured laver 72 chromatography (OPLC) [15] that is in this system fine particle-size of adsorbent layer can be 73 used with forced flow: it is really a planar layer version of HPLC. The exploitation of the unique 74 potential of the planar adsorbent layer for investigation of the mechanism of action of 75 biologically active compounds and for interactions belongs closely to the innovative 76 development of layer liquid system [16].

This review summarizes the progress direction and the results of the development in OPLC
including indirect *in-vitro* biological detection and interactions on adsorbent layers (*in vitro*studies) and partly *in-vivo* biological investigations as well.

80

81 **2. Steps to the development of OPLC**

82 **2.1. Basic elements of ultramicro (UM) chamber and its importance**

83

Attempts to develop an ultramicro (UM) chamber were already made in the 1960s [17]. In this simple chamber, the adsorbent layer on the chromatoplate is covered by a glass plate such a way that the end of the cover plate is not immersed in the eluent. In this way, the failures caused by capillary effect can be avoided. However, because such a rigid glass plate cover did not provide a secure sealing of the adsorbent layer, thus, a UM chamber as shown in Figure 1 is used nowadays. The adsorbent layer is covered with a plastic membrane consisting of a plastic sponge wrapped in plastic foil.

91

92 2.2. Features of pressurized UM chamber: it is the basic instrument of OPLC

93

The further possibilities of UM chamber as a closed planar-layer chamber, such as the increase of eluent-flow velocity, the application of optional development distance, and the use of a pump system for optimizing flow velocity, were realized by the development of a pressurized UM chamber [15,18,19].

98 The essential feature of a pressurized UM chamber system is that the adsorbent layer is 99 completely covered with a flexible membrane under an external pressure so the vapour phase 100 above the adsorbent layer is virtually eliminated. The adsorbent layer is spread onto a flat base 101 plate (preferably in a horizontal position), which is covered by a cushion made from elastic foil or other suitable material, and mounted on the cover plate. The base and cover plates are joined
and the cushion is filled, advantageously with water, making an external pressure. Figure 2
illustrates the main elements of a pressurized circular-type UM chamber [20].

105 The development of the pressurized UM chamber was an original innovative step in the field of 106 LLC [21] using a pump system [22] or other forcing power (e.g. electroosmosis) [23] for the 107 admission of the eluent into the adsorbent laver. This chamber system was the first successful 108 step towards a true planar-layer version of HPLC, as well. The pressurized UM chamber can be 109 regarded as a column: the cushion-like half can be controlled with variable external pressure and 110 the other half forms a rigid plastic or metal plate [24]. The ultramicro character of the 111 conventional and pressurized UM chambers originates from the limited space between the 112 surface of the adsorbent layer and the cover plate and from the space between the adsorbent 113 particles [24].

114

115 **2.3 Development of OPLC instruments and chromatoplates**

116 2.3.1 First commercial OPLC instruments

The essence of OPLC is that the adsorbent layer is closed and pressurized during the separation process and elution can be performed at the highest separation efficiency using the optimum linear velocity generated by a pump. It follows from the principle of OPLC that low (2-5 bar), medium (10-30 bar), and high (50-100 bar or more) external pressures can be used in this forced flow planar layer liquid chromatographic technique.

On the basis of experience gained with experimental pressurized UM chambers, the first commercial pressurized UM chambers were the Chrompres 10 and Chrompres 25 developed in the nineteen eighties by Labor Instrument Works (Budapest, Hungary). The first commercially available OPLC instrument (Chrompres 10) was a completely off-line system [20]. The second generation instrument (Chrompres 25) was suitable for both off-line and on-line separations [25]. The maximum cushion pressure is 2.5 MPa in this chamber and this higher external pressure on the elastic membrane permitted the use of a superfine-particle-size adsorbent layer, higher viscosity mobile phases. At the same time this system makes possible the increase in the eluent front velocity. The main components of Chrompres 25 are shown in Figure 3 [22]. This apparatus is suitable for off-line analytical one-directional, 2-D, and continuous separations and on-line analytical and preparative one-directional linear separations.

Finally, these conventional OPLC instruments and methods were suitable for demonstration of the advantages of OPLC (the progress) over classical TLC and HPTLC [18, 26-29].

135

136 2.3.2 Chromatoplates for conventional OPLC

137

Figure 4 shows the various types of the OPLC chromatoplates. In circular OPLC it is not 138 139 necessary to seal the edges of the chromatoplate and the eluent inlet is placed in the middle of the 140 adsorbent layer. A chromatoplate that is sealed on three sides should be used for one-directional 141 development. If two opposite edges of the chromatoplate are sealed and the eluent inlet is in the 142 middle of the adsorbent layer in a channel, then the system is suitable for a two-directional 143 separation with a large number of samples. For 2-D separation in an off-line system, the four 144 sides of the chromatoplate must be sealed beforehand and the seal opposite the actual inlet must 145 be covered with a strip of filter paper, or an eluent outlet should be used [20, 21, 30-32].

The chromatographic plate for conventional on-line OPLC separation contains two-mobilephase-directing troughs in the adsorbent layer or in the PTFE insert cover plate. The mobile phase inlet trough directs the mobile phase along a linear front and the mobile phase outlet trough collects the mobile phase at the end of the plate and is connected to detector. The combination of several chromatoplates during a single conventional OPLC separation has special advantages (e.g. chromatography of a large number of samples) [20, 21, 33]. The introduction of the mobile phase to parallel-coupled [33] and, especially, serially coupled [28, 34] multi-layer systems is a critical matter. The parallel-coupled is performed by making a perforation in the chromatoplates, of a suitable size and shape, at the mobile phase inlet. For serially coupled multilayer systems, however, a route for transfer of the effluent mobile phase is also needed [28, 34].

156

157 2.3.3 Automatic OPLC instrument and other developments

158

159 On the basis of experience gained with conventional Chrompres chambers, OPLC-NIT Ltd 160 (Budapest, Hungary) developed the automated personal OPLC basic system which includes a 161 separation chamber and a liquid delivery system (Figure 5) [35-37]. The separation chamber 162 generates four main units: holding unit, hydraulic unit, tray-like layer cassette, and attached drain 163 valve. The microprocessor-controlled liquid delivery system includes a two-in-one hydraulic 164 pump and mobile-phase delivery pump. All conditions for single or repeated development can be 165 input and stored in the delivery system software. External pressure (max. 5.0 MPa), mobile phase 166 volume, and mobile phase flow rate can be input, and development time is calculated 167 automatically [36-37].

In automatic OPLC system, the parameters can be adjusted for isocratic and stepwise gradient separations: external pressure (P_{ext}), eluent flow rate (FLR), rapid volume (R or Vol. R), eluent

170 volume A (A or Vol. A), eluent volume B (B or Vol. B), and eluent volume C (C or Vol. C).

171 The development of the automated OPLC instrument has required the use of a new technology;

172 however, this is perhaps only the first step in this innovative direction.

High-pressure circular OPLC (3.8-4.0 MPa), called high-pressure planar liquid chromatography
(HPPLC), was developed by Kaiser and Rieder [38, 39]. HPPLC exploits basically all the
advantages of the original circular operating mode and uses the experience gained in the field of
circular HPTLC, that is, tries to eliminate the problems of the original HPTLC [10,11].

A special version of OPLC was introduced by Witkiewicz et al. [40], who used gas to apply the
external pressure; the mobile phase was fed to the adsorbent layer by means of a syringe.

OPLC Separation Unit 100 (OSU-100), an experimental instrument, is a chamber with three inlets and three outlets. This built-in hydraulic pump already generates 10 MPa external pressure. This technical solution exploits the advantages of a high external pressure which increases the efficiency on the basis of the theoretical relationship. For on-line sample application and detection one or two injectors and UV-detectors (Liquodet 308, Labor MIM, Budapest, Hungary) were connected [41].

185

186 2.3.4. Cassette systems for automated OPLC

187

An interesting aspect of automated OPLC that all conventional OPLC operating modes can be performed by use of an appropriate cassette [35, 36]. The cassette containing the chromatoplate of modified adsorbent layer with the samples can be inserted in the pressurized chamber (Figure 6). Linear one and two-directional, and two-dimensional off-line and linear one-directional inline separations can be performed by isocratic as well as two- and three-step gradient development.

The introduction of a special double-layer cassette system has enabled the possibility of a relatively long development distance which results in increased spot capacity and better resolution [42].

197 **2.4 Main operating steps in OPLC**

198

With present instrumentation, OPLC is suitable for off- and on-line sample applications,
separation and detection, a combination of off- and on-line separation and detection, and
coupling to other techniques, as it can be seen in Figure 7.

202

203 **2.4.1. Off-line OPLC**

204

In off-line systems, all the principal steps in the chromatographic process such as sample 205 206 application, separation, quantitative evaluation, and isolation are performed as separate 207 operations [30]. Depending on the application, linear, circular (radial) and triangular operating 208 modes can be performed in off-line OPLC and each has its own particular merits. In the linear 209 developing mode, one- and two-directional, and 2-D developments are possible [43]. In off-line 210 OPLC systems, changes in the composition of the eluent provide good opportunities for special separation modes; that is, isocratic and stepwise-gradient modes. In analytical off-line OPLC, 211 212 several samples can be processed in parallel. The technique offers further advantages, such as 213 that only the spots and bands of analytical interest need to be assessed, quantitative evaluation 214 can be repeated with various detection parameters. In preparative off-line OPLC after the 215 development the procedures of drying, scraping off the adsorbent layer, elution and 216 crystallization are similar to conventional TLC methods. However, in preparative off-line OPLC, 217 the resolution is considerably increased and thick, fine-particle adsorbent layers can also be used. 218 Fully off-line OPLC has two operations - infusion and transfusion. In infusion mode the mobile 219 phase is introduced into a totally closed adsorbent layer. The air originally present in the 220 adsorbent layer is continuously compressed during the process, helping to fill up the pores of the 221 particles with the eluent. The infusion process is suitable only for off-line developments without 222 overrun and adsorbent layer sealed at four edges has to be used [36]. The transfusion operation 223 mode corresponds to the original OPLC techniques and enables passage of both air and mobile 224 phase [37]. In this system off-line and on-line operation and their combination are possible.

225

226 **2.4.2 On-line OPLC**

If the eluent outlet of the OPLC chamber is connected to a flow-cell detector, eluted solutes can be detected on-line and fractions can also be collected [25, 44]. The entire chromatographic process can be performed on-line by connecting a loop injector to the eluent inlet and a UV detector to the eluent outlet, in much the same way as in HPLC [30, 45].

The hyphenation of TLC/HPTLC with mass spectrometry (MS) via a TLC-MS interface canprovide valuable analytical results [46-48].

On-line OPLC, a genuine layer chromatographic version of HPLC, is especially suitable for
direct coupling to other chromatographic, electrophoretic and/or spectroscopic techniques (e.g.
OPLC-FTIR, OPLC-MS, OPLC-MALDI MS) [49-51].

Fully on-line hyphenation of a newly developed experimental OPLC separation unit with diodearray detection and mass spectrometry (OPLC-DAD-MS) can be used for analysis of biologically active substances (e.g. xanthine derivatives and green tea leaf extract) [52] and this system can, in future, also be used for metabolite analysis. On-line OPLC separation can also be combined with electrospray mass spectrometric measurement (OPLC 50-ESI-MS) and this combination has been used for detection of glycolipids [53].

On-line OPLC-RD (radioactivity detection), and OPLC-DAR (digital autoradiography) and HPLC-RD are rapid, economic and effective separation and detection systems which are highly applicable to animal and human metabolism research (Figure 8) [54].

246

227

247 **3.** Some theoretical aspects of OPLC

248

249 The theoretical aspects of OPLC are summarized in different reviews in more details [32, 36, 37,

250 55-58]; therefore, here in this paper we want to demonstrate only some characteristic theoretical

aspects of OPLC.

252

3.1. Retention and its influencing factors

254

In conventional layer chromatography [TLC, HPTLC and preparative layer chromatography (PLC)], the eluent migrates by means of capillary forces, described by the quadratic equation [15, 18, 19].

In OPLC, the eluent can be forced through (or into in the case of infusion operation) the adsorbent bed by means of a pump system with a selected flow rate. Feeding the eluent by constant flow rate onto the chromatoplate, the speed of the α front (F α) depends on the free cross-sectional area of the adsorbent layer in the direction of the development. Only linear development is able to result in constant linear velocity, chromatoplates with circular and triangular shapes are not. In contrast with linear development circular OPLC yields decreasing velocity of F α along the radius [59].

265

3.2 Relationship between the average theoretical plate height (H) and migration distance (L)

268

Figure 9 shows clearly that H (HETP) is practically constant in off-line OPLC along the adsorbent layer, is independent on development distance. This is more favourable for conventional fine-particle adsorbent layers than for coarse-particle layers [13, 60]. The use of a superfine-particle-size (e.g. 3 μ m) adsorbent layer results in a dramatic increase in the efficiency of the separation; such stationary phases cannot, however, be used in conventional TLC or HPTLC. This arises from the possibility of optimizing the mobile phase velocity in OPLC, by using a pump to direct the mobile phase into the stationary phase. The consequence of the optimization of linear velocity is that two flow profiles also counterbalance each other whensuperfine-particle adsorbent layers are used [61-63].

However, the Figure 9 illustrates clearly the basic difference between the efficiency of TLC,

279 HPTLC, and off-line OPLC and also explains the flow profile of the mobile phase in OPLC.

280

3.3. Relationship between average theoretical plate height (H) and mobile phase velocity (u) in OPLC

283

Figure 10 shows the H-u relationship for different OPLC instruments, including the automatic personal OPLC instrument [35]. It is apparent that increasing the external pressure results in increased optimum mobile phase front velocity with an increased optimum velocity range.

In OPLC, H value can vary characteristically with linear velocity similarly to HPLC [60, 64, 65].

288 The curves from the plot of H against linear velocity (u) for the different operating modes in

289 OPLC are very similar but H values are different. The lowest H values are obtained for fully off-

line OPLC and the highest for fully on-line OPLC. Between these are the two curves of partialoff-line (or partial on-line) OPLC. The differences among these systems originate from "extra-

column" band broadening; this does not occur in the fully off-line system.

The increase of external pressure reduces H in off-line OPLC. Despite of this, the same increase in efficiency was not observed for on-line OPLC [33].

295

296 4. Multi systems – progress in OPLC

297

For to-day it is clear that the efficiency and attractivity of OPLC technique can be increased dramatically by the use of different multi systems which are coming in this case from the attractivity of the adsorbent layer in multiple mode and from the application of a forced(directed) flow of the eluent in it.

302

303 4.1 Multilayer systems

304

305 Parallel solution of overpressured multilaver chromatography (OPMLC) using two or more 306 chromatoplates is very attractive because a large number of samples (50-100 or more) can be 307 separated during one development process [33]. Serial coupled OPMLC (called "long distance" 308 OPLC) can be used for the increase of the theoretical plate number and resolution alike as 309 elaborated by Botz et al. [34]. Several chromatoplates are placed on top of each other to extend 310 the development distance. The potential of these serial connected adsorbent layers can be 311 increased further by use of different (hetero) stationary phases during one development [66] 312 (Figure 11). The end of the first uppermost chromatoplate has a slit-like perforation to enable the 313 mobile phase to flow to a second adsorbent layer, in which the migration continues until the 314 opposite end of the chromatoplate; there the chromatography can be continued onto a subjacent 315 chromatoplate or the eluent is led away.

316

317 4.2 Multidimensional OPLC

318

For difficult separation problems the application of multidimensional (MD) OPLC is necessary, because the power of one-dimensional chromatography is often inadequate for complete resolution of the components which are present in e.g. complex biological samples [67, 68]. According to Giddings [69] the definition of multidimensional chromatography is as follows - The first condition is that the components of a mixture are subjected to two or more separation steps in which their displacements depend on different factors.

- The second criterion is that when two components are substantially separated in any
 single step, they always remain separated until completion of the separation.
- 327

Principle of classical MD TLC (2D-OPLC) development is as follows: chromatographic development in one direction followed by a second development in a direction perpendicular to the first [70]. The method consists of spotting a sample at the corner of a chromatographic plate and enabling migration of the mobile phase in the first direction. After drying, the second development on the chromatoplate, in a direction at right angles to the first, can be started with another mobile phase.

There is a proposed theoretical model whereby maximum peak capacity could be achieved by use of 3D TLC (OPLC) separation [71-73]. Because of technical problems, until now this idea could not be realized in practice.

Figure 12 - from the different technical efforts - illustrates a special solution proposal in patentlevel of 2D and 3D separations.

339

340 4.3 Multichannel OPLC

341

342 A new general concept has been developed for single-channel and multichannel OPLC 343 separations using a nonsegmented adsorbent layer and a flowing eluent wall (FEW) system for 344 operational segmentation [74]. For FEW system the original hydraulic unit of the OPLC has 345 been changed to a new one which is equipped with two mobile phase inlet connections, one for 346 sample injection and another for the FEW formation, and the outlet can be connected to a flow 347 cell detector and/or a fraction collector (Figure 13). The waste eluent from the FEW formation is 348 collected separated. The experimental four and eight-channel FEW versions are suitable for 349 parallel fully on-line separations [41,74-76].

350 For parallel fully on-line separation the FEW can be used for segmentation of a nonsegmented 351 adsorbent layer, dividing it into active and inactive parts with regard to the separation. Only 352 mobile phase is introduced into the inactive part whereas mobile phase and the sample can be 353 introduced to the active separation part; the unsuitable part of the adsorbent layer is thus 354 excluded from the separation process. The FEW helps eliminate the edge effect of OPLC for 355 single sample injection, and abolishes the band widening in horizontal direction, so the sample 356 mixing effect of neighbouring lanes in multichannel separation processes. The FEW as an 357 innovative technical solution in planar layer liquid chromatography enables real multichannel 358 liquid chromatographic separation on a non-segmented adsorbent layer [74, 76]. Figure 14 359 illustrates an eight-channel FEW version which is an integration of multichannel and 360 multidetection systems.

361

362 5. Analytical and preparative applications of OPLC

363

OPLC have already been used successfully in different fields. The up-to-date results have been summarized in details [21, 22, 24, 30, 32, 36, 37, 77]. Therefore, in this review we will only demonstrate some interesting trends in the applications of technical versions of OPLC.

367

368 **5.1 Analytical applications**

369

370 OPLC ensures a constant and high flow velocity, even for viscous mobile-phase mixtures with 371 poor adsorbent wetting characteristics (e.g. those used on RP chromatoplates). The development 372 time is shorter than in TLC or HPTLC, and so molecular diffusion is reduced substantially in 373 OPLC, producing compact spots with better resolution than in TLC or HPTLC [78-80]. The automated Personal OPLC 50 system with maximum 5 MPa external pressure and tray-like cassette construction has been used for separation of a variety of substances (e.g. ascorbigens, aflatoxins, formaldehyde derivatives etc. [81-85]).

377 Comparison of OPLC with other conventional and new planar layer liquid chromatographic 378 techniques is useful for characterization of the separation potential of these techniques [86-379 88].OPLC with off-line hyphenation have been used for separation. OPLC has already combined 380 with in- and/or ex-situ spectroscopic and spectrometric detection using different types of 381 instrument (Raman, FAB, MALDI MS and GC-MS etc.) [89-93].

The newly developed experimental OPLC Separation Unit 100, generating already 10 MPa external pressure, enables more efficient fully off-line (see Figure 10) and on-line separations than previous generations of OPLC system (Figure 15) [76].

385

386 5.2 Preparative applications

387

Although, preparative exploitation of OPLC is poor, up-to-date preparative applications of
OPLC have been reviewed [36, 94, 95].

390 Similar to analytical OPLC off-line and on-line operating modes can be used in preparative 391 OPLC, too. The on-line OPLC is more effective for preparative applications because the time-392 consuming scraping and elution are omissible [36, 95, 96]. In on-line preparative OPLC there is 393 a possibility for the isolation of purer substances than in off-line preparative OPLC.

On-line sample collection after high-resolution stepwise gradient OPLC separation combined with digital autoradiography (DAR) has been used for isolation of plasma and urine metabolites of ³H or ¹⁴C-radiolabeled deramcyclane (a new anxiolytic substance) [97]. This separation process can be followed by different MS techniques for determination of the structure of the minor and major metabolites. OPLC has been used for semipreparative separation-isolation of bound vitamin C from cabbage
extract on analytical silica gel layers [98] and for isolation of xanthines from tea leaf extract
using 0.5 mm preparative adsorbent layer [35].

402

403 **6. Novel study of biological activity of natural and synthetic compounds**

404

405 6.1 Unique great potential of adsorbent layer liquid system for biological detection and 406 interactions – in-vitro and in-vivo studies

407

408 It must be noted firstly that the adsorbent bed in the column arrangement is not suitable and so it 409 can not be used for biological detection and interactions because the living cells (e.g. bacterial 410 cells) do not grow there and so their detection, etc. is not possible. This fact increases the 411 importance of planar layer liquid chromatographic techniques because, in the future time, 412 biological detection and interactions of separated substances in-situ in the adsorbent layers will 413 be a crucial and indispensable methodological solution for isolation, identification and 414 characterization of new antimicrobials, antineoplastics, biopesticides and others [99,100], as well 415 as for studying fundamental biochemical reactions and mechanisms [100,101].

The well-known direct bioautography, which integrates application of planar layer liquid chromatographic separation and post-chromatographic bioassay, can be regarded as the most efficacious assay for detection of antibiotic-like compounds [102]. Although, direct bioautography is a leading technique in bioautography, nowadays it is already not enough.

420

421 **6.2** In-vitro biological studies – BioArena investigations

423 Although the "antibiotic era" is barely six decades old, nevertheless the mankind is already faced 424 with a global problem of emerging resistance of virtually all pathogens [103]. The tasks are 425 given: among others, such small and large co-factor molecules have to know which play a 426 fundamental role in the antibiotic effect and in the basal and induced disease resistance.

427

428 6.2.1 Biochemical-chemical background of antibiotic experiments

429

According to recent observations formaldehyde (HCHO) is an endogenous component of all biological systems [104,105], mainly in the form of hydroxymethyl group. Therefore, it can be detected and measured in different biological samples using a HCHO capture molecule (e.g. dimedone) for collecting HCHO molecules from a given biological unit [106]. There is a primary HCHO cycle in biological systems in which the formation of S-methyl group of L-methionine from HCHO, originated from natural HCHO generators, and the HCHO-yielding function of Sadenosyl-L-methionine (SAM) are essential components [104, 105].

HCHO can be continuously formed by almost all cells both intracellularly and extracellularly. The term "formaldehydome" means the complete set of HCHO-cycle-mediated and nonmediated HCHO pathways of a given biological unit (Figure 16) [107]. The formaldehydome can interlace the big biological units such as genome, proteome etc., so HCHO is a determining endogenous component in all biological systems.

Endogenous HCHO can interact with – among others – H_2O_2 as another universal small endogenous molecule. In this interaction ${}^{1}O_2$ and excited HCHO (HC*HO) can be formed [108, 109]. Figure 17 illustrates the reaction series supposedly [110], taking into account earlier and more recent observations [108-112]. ${}^{1}O_2$ can oxidize the H₂O molecules endogenously and H₂O₃ can be formed from which, among others, O₃ can be released [111, 112]. This fundamental endogenous reaction can occur in *in vitro* conditions (in chromatographic spots), as well as when 448 using the BioArena system [113, 114]. It is probable that O_3 is generated endogenously from 449 different biochemical reactions in diverse biological systems; however, it seems that HCHO/ O_3 450 is a determining pathway in this complicated system.

These very reactive small molecules – from HCHO to O_3 – may be the crucial factors of the innate (natural) resistance and form a common bridge between the innate (natural) and adaptive (induced) resistance in plant and animal organisms, alike [77, 110].

454 BioArena system as the first basically further development of the direct bioautography (e.g. 455 coordination of operating steps, using aimed series of endogenous and/or exogenous molecules 456 [16,100]) can be used to exploit the potential of direct biological detection [100]. Figure 18 457 shows the basic elements of the main possibilities of BioArena. The development of BioArena 458 system leads to new possibilities in bioassay-guided detection, fractionation and isolation. 459 BioArena integrates the advantages of layer liquid chromatography (ideally, variants of linear 460 OPLC [15, 22]), the basic direct bioautography [115-117], and visual, spectroscopic as well as 461 spectrometric evaluation of chromatographic spots before and after biological detection. This 462 integration utilizes the possibilities of interactions among (microbial) indicator cells, the 463 separated components and different small and large co-factor molecules in-situ in the adsorbent 464 layer (in the spots). The possibility of the addition (dissolution) of such endogenous and/or 465 exogenous molecules (co-factors) into the culture medium is unlimited. This possibility is 466 illustrated by means of samples in investigation of the role of HCHO and O₃ in the antibiotic 467 effect.

468

469 6.2.2. Retardation and promotion of antibiotic effect depending on deprivation and
470 increase of HCHO and/or O₃ level in the chromatographic spots

The effect of Cu(II) ions on the antibiotic effect of trans-resveratrol shows a non-linear 472 473 relationship. Cu(II) ions in the pure form generate (mobilize) HCHO molecules from the 474 microbial cells and bind them, possibly forming a coordination complex. It seems that the Cu(II) 475 ion is a concentration-dependent HCHO-generating and capturing as well as transporting ion 476 [118]. It has been observed that the use of the Cu(II) ions in culture medium dose-dependently 477 reduces or promotes the antibacterial activity of trans-resveratrol [119]. Dissolving small doses 478 of Cu(II) ions, the mobilized HCHO molecules in the given system are divided between the 479 Cu(II) ions and the trans-resveratrol molecules. High doses of Cu(II) ions mobilize (induce) 480 more HCHO molecules for most of trans-resveratrol molecules in the chromatographic spot. This is a typical non-linear effect [119]. 481

Using the aqueous suspension of *Saccharomyces cerevisiae* for the biological detection (Figure 19) it can be seen that the HCHO capturer molecules in the culture medium (and so then in the adsorbent layer) can decrease the antiyeast activity of the trans-resveratrol on the silica gel adsorbent layer, while in the presence of Cu(II) ions as HCHO transporting ions the antiyeast activity of trans-resveratrol of the same amount is increased markedly [120].

487 Using the BioArena system, the HCHO molecules could be captured with well-known endogenous HCHO molecules (L-arginine, glutathione) in the spots of paclitaxel (TaxolTM) on 488 489 the TLC/OPLC adsorbent layer after inoculations (Figure 20) [121]. Capture of HCHO resulted 490 in a dose-dependent decrease in the antibacterial activity of paclitaxel. The antibacterial activity 491 of paclitaxel a well-known antitumor diterpenoid alkaloid in the chromatographic spots can be 492 increased dramatically by dissolving Cu(II) ions as a HCHO mobilizing and carrier ions in the 493 culture medium (Figure 20) [121]. By applying an O₃ scavenger (e.g. Indigo Carmine) this 494 oxidant, as a key reaction product of HCHO, could be detected indirectly in chromatographic 495 spots of paclitaxel. It seems that these small molecules – from HCHO to endogenous O_3 – may

496 be crucial factors of the mechanism of antiproliferative action of paclitaxel including killing of497 bystander cancer cells also.

Figure 21 shows the antibacterial activity of the cinnamic acid. It is supposed that O_3 can be formed in the chromatographic spots of cinnamic acid, which is responsible mainly for its antibacterial effect. Using Indigo Carmine for elimination of O_3 molecules from the chromatographic spots of cinnamic acid it has been established that the antibacterial activity of cinnamic acid decreased really dose-dependently as it can be seen in Figure 21 [113].

503 The main components of thyme (Thymus vulgaris L.) essential oil as thymol, carvacrol, and 504 linalool were not appropriately separated by conventional TLC, while OPLC development 505 applying the same stationary and mobile phase gave almost baseline separation (Figure 22, 506 [122]). The densitometric evaluation also confirmed the better resolution between the thymol and 507 carvacrol peaks in the case of OPLC separation. Because of the characteristic functions of mobile phase flow rate, it has to note that the TLC is more time-consuming than OPLC. 508 509 additionally the use of a longer development distance can improve the resolution only in OPLC 510 separation, where the diffusion does not play so important role like in TLC. The direct 511 bioautographic evaluation showed that each of thymol, carvacrol, and linalool has antibacterial 512 effect against luminescent gene tagged Pseudomonas syringae pv. maculicola. It can be seen that 513 in TLC separation the inhibition zones (lack of emitted light, dark area) of thymol and carvacrol 514 were not segregated, as it was expected according to their separation efficiency.

515 OPLC with on-line detection and fractionation is an ideal solution for the efficient separation and 516 isolation of different substances, like antimicrobial components of a complex biological matrix. 517 Moreover, it is very simple to couple on-line as well as off-line this system with different 518 spectroscopic, spectrometric techniques for characterisation and identification of the prospected 519 isolated substances. The usually easy adaptation of a TLC method to OPLC makes possible the 520 use of OPLC for isolation of compounds previously found active in a TLC-direct bioautographic

521 study, that is, for bioassay-guided isolation. Figure 23 shows the steps of the bioassay-guided 522 isolation and identification of some antibacterial components of 50% aqueous ethanol 523 chamomile (*Matricaria recutica* L.) flower extract, applying OPLC fractionation and off-line 524 coupled GC-MS identification [123].

525

526 **6.3.** New aspects of the characterisation of new ingredients – also *in vivo* investigations

527

528 BioArena can be used e.g. for different biochemical reaction between cells and small or large 529 molecules. It seems that in the future the BioArena system can be used for the 530 biological/biochemical characterization of the potential biologically active substances separated 531 beside the chemical and physical characterization.

According to our preliminary observations the in-vitro observations with the BioArena system can be extended to *in vivo* conditions (e.g. greenhouse experiments). This leads to new, useful utilization of results from planar layer liquid chromatographic separations.

On the basis of experiences with the time- and dose-dependent double immune response of plants [110] a logical step was to extend it to the total Avogadro range (i.e. from 10^{-1} to 10^{-23} mol/L⁻¹) in the case there was e.g. four-day interval between pre-treatment and inoculation in the bean-bean pathogen (*Uromyces phaseoli*) relationship. In this system the trans-resveratrol generated a characteristic time- and dose-dependent quadruple immune response of plants to pathogen cells [124] similar to earlier observations with other molecules [77].

It has to note that by capturing HCHO as potential endogenous molecule under *in vivo* conditions
the immunostimulating activity of inducer (e.g. trans-resveratrol) can be eliminated similar to the
BioArena investigations on the adsorbent layer [16].

544

545 **7.0 Conclusions**

547 The elimination of the vapour phase above the adsorbent layer in layer liquid chromatography 548 was first realized by development of the ultramicro (UM) chamber. However, further 549 possibilities of this type of closed simple planar-layer chamber, such as increased eluent-flow 550 velocity, optional development distance, and the use of a pump system for optimizing flow 551 velocity, were achieved by the development of a pressurized UM chamber. The essential feature 552 of a pressurized UM chamber is that the adsorbent layer is completely covered by a flexible 553 membrane under an external pressure so the vapour phase above the adsorbent layer is virtually 554 eliminated. The development of a pressurized UM chamber using a pump system for the 555 admission of the eluent was the first successful step towards a true planar-layer version of HPLC. 556 The pressurized UM chamber is the basic instrument in overpressured layer chromatography 557 (OPLC) that is in forced-flow planar layer LC. The characteristic features and development 558 potential of the original technique are the basis of the diversity in the development of 559 experimental and commercial instruments (e.g. the automated personal OPLC basic system 50, 560 or OPLC Separation Unit 100 (OSU-100)). The development of the multi systems as multilayer, 561 multidimensional, multichannel and multidetection systems shows the direction of the progress 562 in OPLC.

Biological systems such as microbes or plants contain thousands of constituents and are a valuable source of new biologically active molecules, e.g. antibiotics, antineoplastics and biopesticides. For investigation and isolation of them suitable chemical/biochemical methods and biological assays are important. Model experiments are necessary with complex separation and detection systems at microassay and ultramicroassay levels. It is fact that the adsorbent bed in the column arrangement is not suitable for biological detection because the living cells do not grow there and so their detection etc., is not possible. One solution is the BioArena system, which

570 integrates and exploits the advantages of layer chromatographic separation and direct571 bioautography.

572 On the basis of up-to-date theoretical and practical biological and biochemical results obtained 573 by the use of the BioArena system, the formaldehyde (HCHO) and ozone (O_3) as characteristic 574 key small molecules play a crucial role in the antibiotic effect of most diverse chemical 575 substances. Therefore, it is especially interesting to know and to understand better the function 576 and place of these molecules in biological world. It seems that these molecules are determining 577 factors in the antibiosis and the resistance, alike. In future investigations with BioArena system, 578 it has to take into account that the in vitro results (BioArena studies) can be extended to in vivo 579 conditions.

580

581 Acknowledgements

582

The authors are grateful to *Dr. P.G. Ott* (Plant Protection Institute, Hungarian Academy of
Sciences, Budapest, Hungary) for the cooperation in the field of microbiology; also to *Prof. Dr. É. Szőke* and *Dr. Sz. Szarka* (Department of Pharmacognosy, Faculty of Pharmacy, Semmelweis
University, Budapest, Hungary) for GC-MS analysis.

587 This work was partially supported by OTKA grant No. PD83487.

588

589 **References**

- 590
- 591 [1] A.J.P. Martin, R.L.M. Synge, Biochem. J. 35 (1941) 91.
- 592 [2] A.H. Gordon, A.J.P. Martin, and R.L.M. Synge, Biochem. J. 37 (1943) 86.
- 593 [3] R. Consden, A.H. Gordon and A.J.P. Martin, Biochem. J. 38 (1944) 224.
- 594 [4] J.G. Kirchner and G.J. Keller, J. Am. Chem. Soc. 72 (1950) 1867.

- 595 [5] N.A. Ismailov and M.S. Schraiber, Farmacia (Moscow), 3 (1938) 1.
- 596 [6] N.v. Békésy, Biochem. J. 312 (1942) 100.
- 597 [7] J.G. Kirchner, J.M. Miller, G.J. Keller, Anal. Chem. 23 (1951) 420.
- 598 [8] E. Stahl, G. Schröter, G. Kraft, R. Renz, Pharmazie, 1(1956) 633.
- 599 [9] Cs. Horváth (Editor) High Performance Liquid Chromatography, Advances and 600 Perspectives, Vol. 1, Academic Press, New York, NY, 1973.
- 601 [10] R.E. Kaiser (Editor), Einführung in die Hochleistungs-Dünnschicht602 Chromatographie, Institute for Chromatography, Bad Dürkheim, 1976.
- 603 [11] A. Zlatkis and R.E. Kaiser (Eds) HPTLC High Performance Thin-Layer
 604 Chromatography, Elsevier, Amsterdam, 1977.
- 605 [12] U.A.Th. Brinkman and G. De Vries, in: W. Bertsch, S. Hara, R.E. Kaiser and A.
 606 Zlatkis (Eds)): Instrumental HPTLC, Hüthig, Heidelberg, 1980. p. 39.
- 607 [13] R.E. Kaiser, in: W. Bertsch, S. Hara, R.E. Kaiser and A. Zlatkis (Editors):
 608 Instrumental HPTLC, Hüthig, Heidelberg, 1980. p. 165.
- 609 [14] T. Suzuki, T. Kawada and K. Iwai, J. Chromatogr. 198 (1980) 217.
- 610 [15] E. Tyihák, E. Mincsovics and H. Kalász, J. Chromatogr. 174 (1979) 75.
- 611 [16] E. Tyihák, L. Botz, P.G. Ott, J. Nagy, B. Kocsis, Zs. Király-Véghely, E. Mincsovics,
 612 Chem. Anal. (Warsaw) 48 (2003) 543.
- E. Tyihák cited by E. Tyihák and G. Held, in: Progress in TLC and Related
 Techniques, Vol II, A. Niederwieser and G. Pataki (Eds), Ann. Arbor. Science
 Publishers, Ann Arbor, MI, 1971. p. 183.
- 616 [18] E. Mincsovics, E. Tyihák and H. Kalász, J. Chromatogr. 191 (1980) 292.
- 617 [19] H. Kalász, J. Nagy, E. Tyihák and E. Mincsovics, J. Liquid Chromatogr. 3 (1980)
 618 845.
- E. Tyihák, E. Mincsovics, and F. Körmendi, Hung. Sci. Instrum. 55 (1983) 33.

- 620 [21] E. Tyihák and E. Mincsovics, LC-GC Int. 4 (1986) 24.
- 621 [22] E. Tyihák and E. Mincsovics, J. Planar Chromatogr. 1 (1988) 6.
- [23] D. Nurok, J.M. Koers, A.L. Novotny, M.A. Carmichael, J.J. Kosiba, R.E. Santini,
 G.L. Hawkins, R.W. Replogle, Anal. Chem. 76 (2004) 1690.
- K. Ferenczi-Fodor, E. Mincsovics, and E. Tyihák, in: J. Sherma and B. Fried, (Eds)
 Handbook of Thin Layer Chromatography, Marcell Dekker, Inc. New York, 1991. p.
 155.
- E. Mincsovics, E. Tyihák, A.M. Siouffi, in: E. Tyihak (ed.), Proc. Int. Symp. TLC
 with Special Emphasis on Overpressured Layer Chromatography (OPLC), Labor
 MIM, Budapest, 1986, p. 251.
- 630 [26] P. Bruno, M. Caselli, and A. Traini, J. Planar Chromatogr. 1 (1988) 299.
- 631 [27] W.P.N. Fernando and C.F. Poole, J. Planar Chromatogr. 3 (1990) 352.
- 632 [28] L. Botz, Sz. Nyiredy, and O. Sticher, J. Planar Chromatogr. 4 (1991) 115.
- 633 [29] Gy. Kátay, E. Mincsovics, Gy. Szókán and E. Tyihák, J. Chromatogr. A 764 (1997)
 634 103.
- E. Mincsovics and E. Tyihák, in: F.A.A. Dallas et al. (eds), Recent Advances in ThinLayer Chromatography, Plenum Press, New York, 1988, pp. 57.
- 637 [31] E. Tyihák and E. Mincsovics, J. Planar Chromatogr. 23 (2010) 382.
- 638 [32] E. Mincsovics and E. Tyihak, Nat. Prod. Commun. 6 (2011) 719.
- 639 [33] E. Tyihák, E. Mincsovics and T.J. Székely, J. Chromatogr. 471 (1989) 375.
- 640 [34] L. Botz, Sz. Nyiredy and O. Sticher, J. Planar Chromatogr. 3 (1990) 352.
- 641 [35] E. Mincsovics, M. Garami, L. Kecskés, B. Tapa, Z. Végh, Gy. Kátay, E. Tyihák, J.
 642 AOAC Int. 82 (1999) 587.
- E. Tyihák, E. Mincsovics, in: Sz. Nyiredy (ed.), Planar Chromatography A
 Retrospective View for the Third Millennium, Springer, Budapest, 2001, p. 193.

- E. Mincsovics, K. Ferenczi-Fodor, E. Tyihák, Overpressured Layer Chromatography,
 in: J. Sherma, B. Fried (eds.), Handbook of Thin-Layer Chromatography, 3rd ed.
 Marcell Dekker, New York, NY, 2003, p. 171.
- 648 [38] R.E. Kaiser, R.J. Rieder, in: R.E. Kaiser (ed.) Planar Chromatography, Vol. I, Hüthig,
 649 Heidelberg, 1986, p. 165.
- 650 [39] R.E. Kaiser, Einführung in die HPPLC, Hüthig, Heidelberg, 1987.
- [40] Z. Witkiewicz, M. Mazurek, J. Bladek, J. Planar Chromatogr. 6 (1993) 407.
- 652 [41] E. Mincsovics, J. Planar Chromatogr. 17 (2004) 411.
- E. Tyihák, Gy. Kátay, Zs. Ostorics, and E. Mincsovics, J. Planar Chromatogr. 11
 (1998) 5.
- 655 [43] E. Tyihák, J. Pharm. Biomed. Anal. 5 (1987) 191.
- 656 [44] Sz. Nyiredy, C.A.J. Erdelmeyer, K. Dallenbach-Toelke, K. Nyiredy-Mikita, and O.
 657 Sticher, J. Nat. Products 49 (1986) 885.
- 658 [45] K.L. Rowlen, K.A. Duell, J.P. Avery, J.W. Birks, Anal. Chem. 63 (1991) 575.
- 659 [46] G. Morlock, W. Schwack, Trends Anal. Chem. 10 (2010) 1157.
- 660 [47] G. Morlock, W. Schwack, J. Chromatogr. A 1217 (2010) 6600.
- 661 [48] M. Sajewicz, L. Wojtal, M. Hajnos, M. Waksmundzka-Hajnos, T. Kowalska, J.
 662 Planar Chromatogr. 23 (2010) 270.
- [49] B. Betti, G. Lodi, C. Bighi, G. Chiorboli and S. Coppi, J. Planar Chromatogr. 7
 (1994) 301.
- 665 [50] Sz. Nyiredy and K. Glowniak, in: Sz. Nyiredy (ed.) Planar Chromatography A
 666 Retrospective View for the Third Millennium, Springer, Hungary, 2001, p. 551.
- 667 [51] E. Mincsovics, G. Kátay, P.G. Ott, Z. Király-Véghely, Á.M. Móricz, E. Tyihák,
 668 Chromatographia (Suppl.) 62 (2005) 51.

- E. Mincsovics, K. Pápai, K. Ludányi, Á.Z. Dávid, M. Budai, I. Antal, I. Klebovich, J.
 Planar Chromatogr. 21 (2008) 361.
- 671 [53] W. Chai, C. Leteux, A.M. Lawson, M.S. Stoll, Anal. Chem. 75 (2003) 118.
- 672 [54] I. Klebovich, Gy. Morovjan, I. Hazai, E. Mincsovics, J. Planar Chromatogr. 15 (2002)
 673 404.
- 674 [55] F. Geiss, Fundamentals of Thin-Layer Chromatography, Hüthig, Heidelberg,
 675 Germany, 1987.
- 676 [56] C.F. Poole and S.K. Poole, Chromatography Today, Elsevier, Amsterdam, 1991.
- 677 [57] C.F. Poole, in: Sz. Nyiredy (ed.), Planar Chromatography A Retrospective View for
 678 the Third Millennium, Springer, Budapest, Hungary, 2001, p. 13.
- 679 [58] G. Guiochon, N. Marchetti, K. Mrizik, R.A. Shalliker, J. Chromatogr. A, 1189 (2008)
 680 109.
- 681 [59] E. Tyihák, E. Mincsovics, H. Kalász, and J. Nagy, J. Chromatogr. 211 (1981) 45.
- 682 [60] H.E. Hauck and W. Jost, J. Chromatogr. 262 (1983) 113.
- 683 [61] H. Kalász and M. Báthori, LC-GC Int. 10 (1997) 440.
- 684 [62] H. Kalász and J. Nagy, J. Liquid Chromatogr. 4 (1981) 985.
- E. Tyihák, and E. Mincsovics, in: Sz. Nyiredy (ed.) Proc. Planar Chromatogr. 2000,
 Lillafüred, Hungary, Res. Inst. Med. Plants, Budakalász, 2000, p. 37.
- 687 [64] J.H. Knox, J. Chem. Soc. (1961) 433.
- 688 [65] J.C. Giddings, Anal. Chem. 37 (1965) 60.
- 689 [66] Sz. Nyiredy, In: Sz. Nyiredy (ed.): Planar Chromatography- a Retrospective View for
 690 the Third Millennium, Springer, Budapest, Hungary, 2001, p. 103
- 691 [67] C.F. Poole and S.K. Poole, J. Chromatogr. A 703 (1995) 573.

- [68] Sz. Nyiredy, in: R.E. Kaiser, W. Günther, H. Gunz and G. Wulff (eds.) Dünnschicht
 Chromatographie in memoriam Prof. Dr. Hellmut Jork, InCom Sonderband,
 Düsseldorf, 1996, p. 166.
- 695 [69] J.C. Giddings, in: H.J. Cortes (ed.) Multi-dimensional Chromatography, Marcel
 696 Dekker, New York, 1990, p. 1.
- 697 [70] R. Consden, A.H. Gordon and A.J.P. Martin. Biochem. J. 38 (1944) 224.
- 698 [71] G. Guiochon and A.M. Siouffi, J. Chromatogr. 245 (1982) 1.
- 699 [72] G, Guiochon, M.F. Gonnord, M. Zakaria, L.A. Beaver and A.M. Siouffi,
 700 Chromatographia, 17 (1983) 121.
- 701 [73] L.A. Beaver and G. Guiochon, USA Patent No. 4469601, 1984.
- 702 [74] E. Mincsovics, M. Manach, L. Kecskés, B. Tapa, D. Papillard, E. Tyihák, J. Liq.
 703 Chromatogr. Relat. Technol. 26 (2003) 2611.
- [75] E. Mincsovics, In: Proc. Int. Symp. On Planar Separations, Planar Chromatography,
 2005 Milestones in Instrumental TLC , Siófok, Hungary (Ed. Sz. Nyiredy), RIMP,
 Budakalász, 2005, p. 137.
- 707 [76] E. Mincsovics, J. Planar. Chromatogr. 21 (2008) 97.
- 708 [77] E. Tyihák and E. Mincsovics, J. Planar Chromatogr. 23 (2010) 382.
- 709 [78] E. Tyihák, M. Petro-Turza, K. Kardos, E. Mincsovics, J. Planar Chromatogr. 5 (1992)
 710 376.
- 711 [79] J. Pothier, N. Galand, P. Tvollier, C. Viel, J. Planar Chromatogr. 6 (1993) 220.
- 712 [80] A. Linard, P. Gueset, G. Durand, J. Planar Chromatogr. 6 (1993) 322.
- 713 [81] Á. Kovács, L. Simon-Sarkadi, E., Mincsovics, J. Planar Chromatogr. 11 (1998) 43.
- 714 [82] A. Ludwiczuk, Sz. Nyiredy, T. Wolski, J. Planar Chromatogr. 18 (2005) 104.
- 715 [83] E. Tamburini, T. Bernardi, M. Granini, G. Vaccari, J. Planar Chromatogr. 19 (2006)
- 716

10.

- 717 [84] Á.M. Móricz, Zs. Fatér, K.H. Otta, E. Tyihák, E. Mincsovics, Microchem. J. 85
 718 (2007) 140.
- 719 [85] A. Pelander, D. Backström, I. Ojanperá, J. Chromatogr. B 857 (2007) 337.
- 720 [86] J.K. Rozylo, A. Zabinska, J. Matysiek, A. Niewiadomy, J. Chromatogr. Sci. 40
 721 (2002) 581.
- 722 [87] B. Bagocsi, D. Fábián, A. Lauko, S. Maho, Z. Végh, K. Ferenczi-Fodor, J. Planar
 723 Chromatogr. 15 (2002) 252.
- 724 [88] N. Galand, D. Ernouf, F. Montigny, J. Dollet, J. Pothier, J. Chromatogr. Sci. 42
 725 (2004) 130.
- [89] K. Ludányi, K. Vékey, J. Szúnyog, E. Mincsovics, T. Karancsi, K. Újszászy, K.
 Balogh-Nemes, I. Klebovich, J. AOAC Int. 82 (1999) 231.
- [90] E. Horváth, Gy. Kátay, E. Tyihák, J. Kristóf, A. Rédey, Chromatographia 51 (2000)
 297.
- 730 [91] T.K. Rozylo, R. Siembida, Z.I. Németh, L. Albert, E. Tyihák, Biomed. Chromatogr.
 731 14 (2000) 173.
- 732 [92] A. Pelander, I. Ojhanpera, J. Sistonen, I. Rasanen, E. Vuori, J. Anal. Toxicol. 27
 733 (2003) 226.
- 734 [93] H.S. Yang, M.H. Tang, H.X. Deng, J.L. Yang, Sichuan Da Xue Xue Bao Yi Xue Ban,
 735 38 (2007) 1033.
- 736 [94] Sz. Nyiredy, in: Sz. Nyiredy (ed.) Planar Chromatography A Retrospective View
 737 for the Third Millennium, Springer, Budapest, 2001, p. 177.
- E. Mincsovics, in: M. Waksmundzka-Hajnos, J. Sherma, T. Kowalska (eds.), ThinLayer Chromatography in Phytochemistry, CRC Press, Taylor & Frances Group,
 Boca Raton, 2008, p. 215.
- 741 [96] Sz. Nyiredy, J. Chromatogr. A, 1000 (2003) 985.

- [97] I. Klebovich, E. Mincsovics, J. Szunyog, K. Ludányi, T. Karancsi, K. Újszászi, B.
 Dalmadi Kiss, K. Vékey, J. Planar Chromatogr. 11 (1998) 394.
- 744 [98] E. Mincsovics, É. Sárdi, I. Velich, Gy. Kátay, E. Tyihák, J. Planar Chromatogr. 15
 745 (2002) 280.
- 746 [99] D.E.Wedge, N.D. Camper, in: H.G. Cutler, S.J. Cutler (eds.), Biologically Active
 747 Natural Products: Agrochemicals and Pharmaceuticals, CRC Press, Taylor & Francis
 748 Group, Boca Raton, 2000, p. 1.
- [100] E. Tyihák, Á.M. Móricz, P.G. Ott, in: M. Waksmundzka-Hajnos, J. Sherma, T.
 Kowalska (eds.), Thin-Layer Chromatography in Phytochemistry, CRC Press, Taylor
 and Francis Group, Boca Raton, 2008, p. 193.
- [101] E. Tyihák, P. Ott, Á.M. Móricz, Gy. Kátay, Zs. Király-Véghely, J. Planar
 Chromatogr. 17 (2004) 84.
- 754 [102] J.I. Rios, M.C. Recio, A. Villar, J. Ethnopharmacol. 23 (1988) 127.
- 755 [103] J.N. Eloff, S. Afr. J. Sci. 96 (2000) 116.
- [104] E.Tyihák, L. Albert, Zs.I. Németh, Gy. Kátay, Zs. Király-Véghely, B. Szende, Acta
 Biol. Hung. 49 (1998) 225.
- 758 [105] E. Tyihák, L. Trézl, and B. Szende, Ann. N. Y. Acad. Sci. 851 (1998) 259.
- 759 [106] Gy. Kátay, M. Adrian-Romero, Zs.I. Németh, G. Blunden, E. Tyihák, L. Albert, J.
 760 Planar Chromatogr. 15 (2002) 28.
- [107] E. Tyihák, E. Mincsovics, Gy. Kátay, Zs. Király-Véghely, Á.M. Móricz, P.G. Ott, J.
 Planar Chromatogr. 21 (2008) 15.
- 763 [108] L. Trézl, J. Pipek, J. Mol. Struct. Theochem. 170 (1988) 213.
- [109] E. Tyihák, Zs. Rozsnyay, É. Sárdi, G. Gullner, L. Trézl, R. Gáborjányi, Acta Biol.
 Acad. Sci. Hung. 45 (1994) 3.

766	[110]	E. Tyihák, in: J.A. Teixeira da Silva (ed.) Foliculture, Oriental and Plant
767		Biotechnology, Advances and Topical Issues Vol 3, Global Science Books, London,
768		UK, 2006, p.380.

- 769 [111] A.D. Wentworth, I.H. Jones, P. Wentworth, K.D. Janda, R.A. Lerner, Proc. Natl.
 770 Acad. Sci. USA 97 (2000) 10930.
- [112] P. Wentworth Jr., A.D. Wentworth, X.Y. Zhu, L.A. Wilson, K.D. Janda, A.
 Eschenmoser, R.A. Lerner, Proc. Natl. Acad. Sci. USA 100 (2003) 1490.
- 773 [113] E. Tyihák, Á.M. Móricz, P.G. Ott, J. Planar Chromatogr. 21 (2008) 77.
- 774 [114] Á.M. Móricz, N. Adányi, E. Horváth, P.G. Ott, E. Tyihák, J. Planar Chromatogr. 21
 775 (2008) 417.
- 776 [115] R.R. Goodall and A.A. Levi, Nature 158 (1946) 675.
- 777 [116] Á.M. Móricz, P.G. Ott, F. Billes, K.-H. Otta, E. Tyihák, J. Appl. Microbiol. 103
 778 (2007) 2525.
- 779 [117] I.M. Choma, E.M. Grzelak, J. Chromatogr. A, 1218 (2011) 2684.
- [118] E. Tyihák, A. Takátsy, Á.M. Móricz, P.G. Ott, R. Ohmacht, in: M. Szilágyi and K.
 Szentmihályi (eds), Proc. Trace Elements in the Food Chain, Vol.3, Budapest,
 Hungary, 2009, p. 292.
- [119] E. Tyihák, Á.M. Móricz, P.G. Ott, Gy. Kátay, and Zs. Király-Véghely, J. Planar
 Chromatogr. 18 (2005) 67.
- [120] E. Tyihák, Gy. Kátay, Á.M. Móricz, P.G. Ott, Zs. Király-Véghely, Acta Pharm.
 Hung. 77 (2007) 53.
- [121] E. Tyihák, Á.M. Móricz, P.G. Ott, M.L. Hajnos, and K. Glowniak, J. Planar
 Chromatogr. 21 (2008) 331.
- [122] Á.M. Móricz, Gy. Horváth, P. Molnár, B. Kocsis, A. Böszörményi, É. Lemberkovics,
 P.G. Ott, J. Planar Chromatogr. 23 (2010) 406.

791	[123]	Á.M.	Móricz,	Sz.	Szarka,	P.G.	Ott,	É.B.	Héthelyi,	É.	Szőke,	E.	Tyihák,	Med.
792		Chem	.(in Press	5)										

793 [124] E. Tyihák, Zs. Király-Véghely, and Á.M. Móricz, Nat. Prod. Commun. 6 (2011) 631



799 Schematic drawing of an ultramicro chamber

800 Distances are in millimetres [20]



- 801
- 802 Figure 2
- 803 Schematic drawing of the circular type of pressurized ultramicro chamber

A, water inlet, B, developing solvent inlet, C, pressure gauge, D, screw fastener, E, rubber Oring, F, adsorbent layer, G, support plate, H, plastic foil cushion system, J, polymethacrylate support blocks [20]



808 Figure 3

807

809 Schematic drawing of Chrompres 25

A, chamber, B, eluent pump, C, safety valve, D, pressure gauge, E, solvent-inlet valve, F, water outlet, G, polymethacrylate cover, H, solvent-outlet pin, J, water inlet, K, water pump (with permission, from Ref. [22]).



813

814 Figure 4

Schematic drawing of chromatographic chromatoplates used in conventional OPLC separations;
a, one-directional; b, two-directional; c, circular; d, two-directional; c, on-line; f, parallel
coupled multilayer; g, serial coupled multilayer; h, parallel-serial coupled multilayer. (With
permission, from Ref. [36].)









822 1, liquid-delivery system; 2, separation chamber; 3, cassette; 4, mobile phase inlet; 5 mobile

823 phase outlet; 6, mobile phase switching valve; 7, mobile phase reservoirs; 8, LCD display. (With

824 permission, from Ref. [41]).



- 825
- 826 Figure 6
- 827 Cassette solutions to automatic OPLC
- 828 al, cassette for system rinsing
- 829 a2, cassette for bidirectional infusion development of 20x20 cm layer
- b, cassette of 20x20 cm layer for linear one- and two-dimensional development
- c, cassette of 10x20 cm glass backed layer for linear one-dimensional development;
- 832 d, cassette for circular development.
- 833 Arrow represents the eluent movement.
- 834 1, PTFE coverplate involves eluent directing trough and hole is in floating position; 2, eluent-
- 835 directing trough; 3, sample; 4, adsorbent layer
- 836



- 838 Figure 7
- 839 Schematic diagram of off-line and on-line OPLC. Dashed arrows show the procedure of off-line
- 840 operating mode.

ISOLATION OF ¹⁴C LABELLED METABOLITES BY COMBINED OPLC-HPLC SYSTEM



842 Figure 8

- 843 Isolation of ¹⁴C labelled metabolites by combined OPLC-HPLC system
- 844 Normal phase OPLC separation (left) reversed phase HPLC separation (right).
- 845 Peak marked by asterisk has been separated for isolation.





Figure 9

848 Correlation between the average theoretical plate height (H) and the distance (X) travelled by the

849 mobile phase on silica gel layers of different particle size and in different chamber systems.

850 Ns, normal saturated chamber; Nus, normal unsaturated chamber; UM, ultramicro chamber

851 (totally closed adsorbent layer, without overpressure); OPLC chamber using 11, 5 and 3 μm
852 particle size, respectively (with permission from Ref. [36]).







Effect of linear velocity (u) on theoretical plate height (H) at different external pressures. The results were obtained from the use of the experimental OPLC Separation Unit 100 with the following conditions:

 $5 \ \mu$ l of a 0.36 mg mL⁻¹ PTH-valine solution was applied in 10 mm band, and chromatographed on HPTLC silica gel (Merck) with dichloromethane-ethyl acetate 92:8 (v/v) as mobile phase; the development distance was 170 mm. (With permission, from Ref. [77]).



861

Schematic diagram of multilayer (ML)-OPLC using different stationary phases of decreasing
polarity. (a) Arrangement for analytical off-line separation of eighteen samples or one sample for
micropreparative separation. (b) Arrangement for on–line analytical or micropreparative
separation.

Figure 11





868 Figure 12

- 869 Schematic drawing of 1-, 2-, and 3-dimensional OPLC separation.
- 870 (Figure is from a patent application, and lecture in Aix-an Provence, 1992.)





873 Schematic diagram of a four-channel OPLC system with the FEW-I/O arrangement dedicated to 874 fully on-line separation of four samples (a), two samples (b), or one sample (c) using 20 x 20 cm, 875 and two samples using 10 x 20 cm (d) adsorbent layers. 1, non-segmented adsorbent layer; 2, 876 sealed edges; 3, distributor space of mobile phase for FEW formation; 4, distributor space for 877 sample application; 5, FEW part of adsorbent layer; 6, separation part of adsorbent layer; 7, 878 components separated; 8, sample collector space at outlet side; 9, collector space of FEW at 879 outlet side; E, mobile phase; I, injectors; D, detectors; W, waste collected by FEW lines (with 880 permission, from Ref. [76).



882 Figure 14

881

883 Schematic diagram of eight-channel OPLC separations using parallel injections and FEW-I/O

arrangements. Further conditions are at Figure 13.





887 One-channel fully on-line FEW-OPLC separation of some PTH-amino acids using the 888 experimental OPLC Separation Unit 100. The acids were separated on a 5 cm x 20 cm fine-889 particle silica gel layer with chloroform-ethyl acetate 9:1 (v/v) as mobile phase at a flow rate of 890 0.3 mL min⁻¹; injection volume, 0.5 μ l; cell volume , 1 μ l; detection at 275 nm; AUFS 0.05/0.5V. 891 1, PTH, proline; 2, PTH-leucine; 3, PTH-isoleucine; 4, PTH-valine; 5, PTH-methionine; 6, PTH-

892 glycine (with permission, from Ref. [41]).



894 Figure 16

893

895 The main elements of the formaldehydome (with permission, from Ref. [77]).



- 896
- 897 Figure 17

898 Possible formation of different reactive oxidants by interaction of HCHO and H₂O₂ (with

899 permission, from Ref. [77]).



900

901 Figure 18

- 902 Main steps in the detection and identification of a typical spot with the BioArena system (with
- 903 permission, from Ref. [77]).



905 Figure 19

904

The influence of endogenous substances on antiyeast (*Saccharomyces cerevisiae*) activity of trans-resveratrol. Chromatographic conditions: silica gel 60 F_{254} (Merck, preconditioning at 120 °C for 3 h), chloroform-methanol 80:4 (v/v). Biological conditions: (A) yeast suspension (3 g yeast in 100 mL distilled water); (B) A + 2 mg L-arginine in 1 mL yeast suspension; (C) A + 2 mg reduced glutathione in 1 mL yeast suspension; (D) A + 4 mg CuSO₄ x 5 H₂O in 100 mL yeast suspension. (With permission, from Ref. [120]).



913 Figure 20

The influence of endogenous substances on the antibacterial activity of paclitaxel. Chromatographic conditions: silica gel 60 F_{254} (Merck); mobile phase: chloroform-methanol 90:10 (v/v); Desaga glass TLC chamber. Biological conditions: Control: *Pseudomonas* sp., other information is on the chromatograms. Incubation was at the optimum temperature and detection was performed with MTT. (With permission, from Ref. [121].)



920 Figure 21

921 Effect of Indigo Carmine on the antibacterial activity of cinnamic acid. Chromatographic 922 conditions: silica gel 60 F_{254} (Merck); mobile phase: chloroform-methanol 80:8 (v/v); Desaga 923 glass TLC chamber. Biological conditions: (A) 100 mL *Pseudomonas savastanoi* suspension, 924 (Psm); (B) A + 10 mg Indigo Carmine; (C) A + 20 mg Indigo Carmine; (D) A + 40 mg Indigo 925 Carmine.







Detection of the main components of *Thymus vulgaris* L. essential oil separated by TLC (I) and OPLC (II). Track 1, standards, 10 μ g each of thymol (A), carvacrol (B), and linalool (C). Track 2, *Thymus vulgaris* L. oil, 25 μ g. The components were detected before biological detection by densitometry (a – track 1, b – track 2) and under UV light (254 nm) (c); The bioautograms (d) were obtained by use of luminescent *Pseudomonas syringae* pv. *maculicola*. (With permission, from Ref. [122].)







Bioassay-guided isolation and identification of antibacterial components of 50% aqueous ethanol
extract of chamomile flower (150 mg in 1 mL);

A – The active components, separated on TLC layer (Merck, #5554) with chloroform-acetone
999:1 (v/v), were visualized under UV (254 nm) (on the left) and with direct bioautography using
luminescent gene tagged *Pseudomonas savastanoi* pv. *maculicola* plant pathogen bacteria (on
the right);

B – OPLC separation of 0.5 mL chamomile flower extract, performed by a Personal OPLC BS50
system (OPLC-NIT, Budapest, Hungary), and the collected fractions (1-3, the 3 peaks); OPLC
conditions were as follows: sealed 20x20 cm normal particle size silica gel layer (Merck, #5554),
sample application in a 16 cm wide band at 3 cm from the edge, 50 bar external pressure, 450 μL

- rapid mobile-phase flush, 1 mL min⁻¹ mobile phase flow rate; the mobile-phase was chloroform,
 the on-line UV detection was achieved at 350 nm;
- 948 C The TLC re-chromatography of the collected fractions (chloroform-acetone 99:1 (v/v)); the 949 active components were visualized under UV (254 nm) (on the left) and with direct 950 bioautography using luminescent gene tagged *Pseudomonas savastanoi* pv. *maculicola* plant 951 pathogen bacteria (on the right);
- 952 D GC-MS analyses of the 3 fractions (TIC chromatograms);
- 953 E The MS spectra of the main components of the chamomile flower fractions;
- 954 F Chemical structures of identified components: a *cis*-spiroether, b *trans*-spiroether, c –
- 955 herniarin [123].
- 956
- 957