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Stable isotope compositions of bivalve shells and geochemistry of bulk sediments in a 5–20 ky fluvial section at Körösladány, SE Hungary: Sedimentary changes vs. climate signals

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In this paper we present sedimentological and geochemical data for a section of fluvial deposits from SE Hungary covering the period from about 20 to 5 ky BP. Major and trace element geochemistry of bulk sediments as well as stable C and O isotope compositions of the carbonate content indicate significant changes in depositional facies and/or sediment provenance as well as climate conditions. Variations in bulk sediment Sr, TiO2 and P2O5 concentrations were correlated with major climate change events following the Late Glacial Maximum that support the age model established on the basis of AMS ¹⁴C age data. Bulk sediment Sr concentrations and stable C and O isotope compositions of bulk sediment carbonate were determined by changes in denudation of carbonate rocks in the recharge area. The Sr and C–O isotope patterns show correlations with global temperature changes during the Pleistocene-Holocene transition. However, TiO2 and P2O5 contents show correspondence with humidity changes, suggesting variations in chemical weathering. In addition to the sedimentological effects, C and O isotope compositions of Unio crassus shell fragments show strong changes at the Pleistocene-Holocene transition, indicating that the bivalve shells can reflect climate conditions. On the other hand, shorter climate change events were difficult to track in the isotope records due to the competing fractionation processes. The combined evaluation of chemical and isotopic compositions revealed that beside the globally important Younger Dryas and Bølling/Allerød periods, the Ságvár-Lascaux interstadial was of local importance, in accordance with earlier studies.

Key words: Pleistocene-Holocene transition, Late Glacial, stable isotopes, geochemistry, fluvial sediments, *Unio crassus, Sphaerium rivicola*

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Introduction

Quaternary climate change events have been well documented in ice core (e.g. Petit et al. 1999; EPICA community members 2004) and deep sea carbonate sediment stable isotope records (Imbrie et al. 1984). Large-scale changes (10 to 100 thousand years [ky]) are attributed to variations in the Earth's orbital parameters (see EPICA community members 2004, as a comprehensive study), whereas climate fluctuations on the millennial to decadal scale are interpreted as results of a complex interplay of insolation, greenhouse gas concentration, ocean water circulation and hydrosphere-atmosphere interactions (e.g. Bond et al. 1999, 2001; Paillard 2001; Versteegh 2005; Schulz and Zeebe 2006; Denton et al. 2010; Hurrell and Deser 2010; Morley et al. 2011). Beside marine and ice core sections, terrestrial deposits can also provide useful information on paleoclimate conditions. Loess sections have been successfully used to detect and correlate climate change events in the last 100 ky in Europe (Shi et al. 2003). Within this period, the last 20 ky is of special significance, as maximum glaciation as well as warming to the present interglacial are included. Several climate fluctuation events have been distinguished within this period, including sudden cooling events (Heinrich Stadial [H1] and Younger Dryas [YD]), and relatively warm periods like the Bølling/Allerød. The Holocene warming began after the Younger Dryas period at 11700 years cal BP. These climate fluctuation events can be globally detected and correlated (Shakun and Carlson 2010). A number of studies have dealt with late glacial climate change events in the Carpathians and the Carpathian Basin on the basis of age dating, sedimentological, paleobiological and geochemical analyses conducted on loess sequences and lake deposits (e.g. Sümegi and Hertelendi 1998; Sümegi and Krolopp 2002; Feurdean et al. 2008; Novothny et al. 2009; 2011; Sümegi et al. 2011a; Schatz et al. 2011, 2012; Buczkó et al. 2012; Magyari et al. 2012a; Tóth et al. 2012). Although the results of the different paleoenvironmental reconstructions are sometimes conflicting, climate variations characterized by alternations of colder and warmer, arid and humid or windy and less windy conditions have been determined within the period of 12-80 ky BP (Sümegi and Rudner 2001; Schatz et al. 2011; Stevens et al. 2011). Speleothems that provide globally and regionally important paleoclimate data are absent from the list of late glacial paleoclimate records of the Carpathian region as stalagmite formation generally ceased in this period, due to the cold and dry climate (Siklósy et al. 2011).

A number of studies have established temperature and humidity changes in the late glacial period. The temperature change related to the Pleistocene-Holocene transition was at least 4.9°C in terms of global mean, whereas the rise in annual mean temperature within the Carpathian Basin reached about 8–10°C on the basis of malacothermometry (Sümegi and Krolopp 2002), noble gas thermometric groundwater analyses (Varsányi et al. 2011) and stable oxygen isotope compositions of fossil teeth (Kovács et al. 2012). Using fossil chironomid species analyses on lake sediments from the Southern Carpathians, Tóth et al. (2012) determined mean July air temperature variations for the latest glacial and the Pleistocene-Holocene transition. They obtained an $\sim 8^{\circ}$ C rise from the glacial maximum to the Holocene, with the Younger Dryas event showing only a slight (<1°C) decrease. Since the Younger Dryas is thought to represent strong cooling in the northern hemisphere (Shakun and Carlson 2010), Tóth et al. (2012) interpreted this discrepancy as a sign of increased seasonality with very cold winters.

Beside temperature, another important climate factor is humidity, i.e. precipitation amount. Not only the mean annual precipitation, but also precipitation seasonality can be determined on the basis of paleobiological proxies and lake level estimations, as exemplified by a number of studies conducted in the Carpathian region (e.g. Tullner and Cserny 2003; Jenei et al. 2007; Sümegi et al. 2011b; Magyari et al. 2009, 2012a). Recent studies have shown that even the variations in precipitation origin, i.e. whether the moisture derived from Atlantic or Mediterranean transport trajectories, can be determined on the basis of stable isotope analyses of lake sediments and speleothems (Demény et al. 2012; Magyari et al. 2012b).

Beside the proxies described above, a further possibility to study climate conditions in the warm season is provided by freshwater bivalve shells, whose stable isotope compositions indicate temperature and humidity variations (Schöll-Barna et al. 2012 and references therein). For Lake Balaton the climate - shell composition relationship is well established (Schöll-Barna et al. 2012), while the fluvial environments of the Carpathian Basin have not been studied. The aim of the present study was to determine how a major climate change event, the Pleistocene-Holocene transition, is reflected by Unio shell isotope compositions, and how the smaller-scale fluctuations in the late glacial period can be traced by these data. A 5.2 m-thick fluvial section was studied in this work to detect climate signals during the late Pleistocene and early Holocene. The sedimentary section was cleaned and collected for paleontological, geologic and geochemical analyses in the clay quarry which is located at the edge of the village of Körösladány in the southeastern part of the Great Hungarian Plain (Hungary) (Fig. 1). Based on preliminary ¹⁴C age dating it was suspected that the section would cover the last approx. 20 ky. Sampling the section at 20–25 cm intervals was conducted, which would thus allow paleoclimate analyses at a millennial age resolution for the study of the main fluctuation processes. Unfortunately the section has been destroyed by the use of the quarry material prohibiting further, higher-resolution studies; thus this paper serves as an archive of available data. Apart from the shell analyses, geochemical analyses of bulk sediment samples were also conducted to determine erosion processes. Additional AMS (Accelerator Mass Spectrometry) ¹⁴C age dating on seven shell samples supplemented the age dataset in order to establish a more reliable age-depth model that was thus based on 10 ages from the \sim 5 m-thick section.



Fig. 1 Location of the studied section

Analytical methods

Grain-size analysis followed the Casagrande-type hydrometric method (Vendel 1959). Lithostratigraphic and lithofaciological description of the sections were carried out according to the system of Troels-Smith (1955) and Bridge (1993). Radiocarbon dating was performed on mussel shells recovered from the section. As a part of a pilot study, conventional ¹⁴C analyses were carried out at the Poznan Radiocarbon Laboratory, the Light Isotope Laboratory of the Nuclear Research Center of the Hungarian Academy of Sciences, and the Gliwice Radiocarbon Laboratory (one age analysis at each laboratory, samples at 100, 225 and 520 cm). Subsequently new accelerator mass spectrometry (AMS) age dating was conducted for this study at the Scottish Universities Environmental Research Centre (SUERC, seven samples distributed throughout the section; see Table 2) to supplement the earlier data and to obtain a more reliable age-depth relationship. The preparation of the samples and the actual steps of the measurement followed the methods of Hertelendi et al. (1989, 1992). Measurements at the SUERC were conducted using a single-stage accelerator mass spectrometer (Freeman et al. 2010) and radiocarbon ages were calculated using the background subtraction method. Subsequently, all the radiocarbon dates were calibrated using the University of Oxford Radiocarbon Accelerator Unit calibration program (OxCal4). The original dates are indicated as yr BP, whereas the calibrated dates are indicated as cal BP years.

All of the following mineralogical and geochemical analyses were conducted at the Institute for Geological and Geochemical Research, Budapest. Bulk sediment samples were milled in an agate mortar to fine powder ($<50 \mu$ m), followed by analyses of mineralogical compositions by X-ray diffractometry (Philips PW 1730 type diffractometer). Trace element contents were determined on powder tablets using a Philips 1410 X-ray fluorescence spectrometer.

Fragments (usually 2–3 cm in size) of *Unio crassus* and *Sphaerium rivicola* shells were collected from the sediments at 20–25 cm intervals and the preservation of their original aragonite structure was checked by cathodoluminescence

microscopy, applying a Reliotron-type cold-cathode equipment attached to a Nikon Eclipse E600 optical microscope with a Nikon Coolpix 4500 digital camera.

Carbon and oxygen isotope compositions of bulk sediment carbonate were determined using the conventional H_3PO_4 digestion method at 25 °C (McCrea 1950), but applying a 2-day reaction time due to the dolomite content. The $^{13}C/^{12}C$ and $^{18}O/^{16}O$ ratios of CO_2 generated were measured with a Finnigan MAT delta S mass spectrometer. Bivalve shells were sampled on their outer surface by using a 1-mm drill bit, collecting carbonate powder at three drilling points for each shell fragment. Batches of 0.1–0.2 mg of carbonate were analyzed using an automated GASBENCH attached to a Finnigan Thermo delta plus XP mass spectrometer (see Spöt1 and Vennemann 2003); thereafter the average values for individual fragments were calculated and used in this study. Standardization was conducted using laboratory calcite standards calibrated against the NBS-19 standard. The results are expressed in the δ -notation in % relative V-PDB [$\delta = (R_1/R_2-1) \times 1000$] where R_1 is the $^{13}C/^{12}C$ or $^{18}O/^{16}O$ ratio in the sample and R2 the corresponding ratio of the standard. Reproducibility for both was better than $\pm 0.15\%$.

Results

Sedimentology

According to the sedimentological, geomorphological and geofaciological investigations, the analyzed section (summarized in Table 1) consists of the sediments of a filled-up point bar channel.

The basal sediment layer consists of fluvial very fine, fine and medium-grained sand; at the center of the paleo-point bar channel the basal sandy material contains significant amounts of mussel (Unio) fragments and iron-hydroxide spots and noodles (550–480 cm). At the top of the basal fluvial layer a silt-rich,

Table 1

Lithological and layer changes in the analysed profile of the brickyard at Körösladány

Depth (cm)	Troels– Smith description	Primary layers structure	Secondary layers structure	Description
280–0	Ag2As1Lc1	Fine parallel laminated	Iron hydroxide and carbonate noodles – spots	Fine alluvial sediment within eolian dust (infusion loess)
300-280	Ga4	Cross-bedding	_	
330-300	Gs1Ga3	Wavy laminated	—	Active fluvial layer
400-330	Ga4	Cross-bedding	-	within the cyclically
420400	Gs1Ga3	Wavy laminated	Iron hydroxide noodles – spots	sedimentological content changes
450-420	Ga4	Cross-bedding	-	
480-450	As2Ag2	Fine parallel laminated	_	Fine alluvial sediment layer
550-480	Gs1Ga3	Cross-bedding	Iron hydroxide noodles – spots	Mussel rich sandy fluvial layer

parallel-laminated alluvial layer formed (480-450 cm), which suggests that the transportation energy of the analyzed river declined. The next unit consists of different sandy layers (450–280 cm). Grain-size compositions, together with the sedimentary features and macrostructure, indicate cyclically-decreasing and increasing flood intensity in this period. The coarser sandy and wavy laminated layers (420–400 cm, 330–300 cm) suggest that the accumulation energy increased in the point bar channel, while the finer sandy and cross-bedding laminated ones (450-420 cm, 400-330 cm, 280-300 cm) formed under lower accumulation energy conditions. This cyclic sedimentation change suggests that the energy of the river and flood intensity changed gradually in this period for either tectonic and/or climatic reasons. At the top of this fluvial sediment sequence an infusion loesslike, silt-rich alluvial sediment layer formed within carbonate and iron-hydroxide noodles and spots. The development of this layer indicates that the active river phase ended in this region and alluvial deposits accumulated in the point-bar channel system. The presence of some terrestrial warmth-loving mollusk species, such as Cepaea vindobonensis, Chondrula tridens, and Granaria frumentum, suggest that this alluvial sedimentation developed in the postglacial phase, during the Early Holocene. Finally, the uppermost 50 cm is disturbed by anthropogenic influence, and soil formation is above 80 cm. Although the effect of soil formation does not necessarily affect the chemical compositions of the sediment, it is indeed detected in the preservation of *Unio* shells that are calcitized to different degrees in the uppermost part of the section. However, based on cathodoluminescence microscopic analyses, pure aragonite is preserved in the shells at and below 80 cm.

Geochemistry

Age-depth model calculations were carried out using the StalAge algorithm of Scholz and Hoffmann (2011) based on selected age dates (see Table 2 and Fig. 2). Ages at 150 and 475 cm depths were not in stratigraphic order, whereas the age obtained at 275 cm is an outlier compared to the neighboring ones (with an

Sample	Depth (cm)	Lab code	Years BP	Error	Cal yr BC
KLM0	520	GdA-557	21490	110	24015-23569
KLM2	475	SUERC-36616	13975	45	15175-14957
KLM4	425	SUERC-36617	14155	45	15315-15093
KLM6	375	SUERC-36618	13335	45	14763-14332
KLM8	325	SUERC-36619	12925	40	13663-13199
KLM10	275	SUERC-36620	13740	45	14990-14829
KLM12	225	deb-1069	11700	250	11851-11364
KLM15	150	SUERC-36621	5905	40	4801-4724
KLM17	100	Poz-23316	6890	40	5809-5726
KLM18	80	SUERC-36625	5970	40	4857-4796

Table 2 ¹⁴C age dating data (68% range) determined for *Unio crassus* shells



Age-depth model (solid black line) with 95% confidence limits (grey lines) for the Körösladány section based on the StalAge algorithm of Scholz and Hoffmann (2011). Samples excluded from the calculation (see text) are marked by empty circles

unexpectedly older age that may be related to re-sedimentation of shell fragments); hence these data were excluded from the age-depth model (see Fig. 2).

The mineralogical composition data are listed in Table 3. As usual for fluvial deposits, the studied sediments are dominated by quartz (54–68 wt.%) and plagioclase (8–20 wt.%), while phyllosilicates, feldspars, chlorite and carbonates are generally below 10 wt.%. Between 300 and 375 cm the calcite content rises to 10–13 wt.% with a concomitant increase of dolomite to 5–9 wt.%. The samples contained no aragonite. This is partly due to the collection of large shell fragments from the sediment samples. Below 400 cm (>17 ky cal BP), the carbonate amount is low with a relatively elevated calcite/dolomite ratio (expressed as Xcalcite in the bulk carbonate, Fig. 3A). Going up-section, the dolomite content suddenly increases at 375 cm (~164 ky BP), then shows a constant decrease along with the calcite/dolomite ratio (Fig. 3B).

Bulk sediment chemical compositions also display systematic variations (Table 4). The SiO₂ content is slightly elevated (57 to 69 wt. %) below 400 cm (~17 ky BP)

Table 3 Mineralogical compositions (in wt.%) of bulk sediment samples from the Körösladány section

Depth (cm)	Qu	Pl	Kfp	Shs	Chl	Cc	Dol
25	65	13	4	5	3	4	5
50	67	9	6	5	4	4	6
75	62	12	7	5	3	6	5
100	64	12	3	6	3	6	5
125	66	11	2	5	4	6	4
150	63	13	4	5	4	7	5
175	67	13	3	5	3	4	5
200	55	20	4	5	3	7	7
225	63	12	3	5	3	7	6
250	61	14	4	5	3	9	5
275	56	14	4	6	4	9	7
300	54	13	5	5	3	13	8
325	57	12	3	6	5	10	5
350	58	10	3	4	3	13	9
375	66	8	4	4	5	8	5
400	68	14	2	5	3	6	1
425	66	15	4	5	2	6	1
450	67	15	4	5	3	5	1
475	65	17	3	5	2	6	1
500	62	13	3	5	4	11	2

Qu: quartz, Pl: plagioclase, Kfp: K-feldspar, Shs: 10 Å sheet silicates, Chl: chlorite, Cc: calcite, Dol: dolomite

compared to the upper section part (52 to 62 wt.%) (Fig. 3C). The MgO concentration is low below 400 cm (1.7 to 1.9 wt.%), suddenly increases to 34 wt.%, then decreases gradually to about 2.4 wt.%, with a significant positive MgO-dolomite correlation (R^2 =0.67). The MnO content shows again a difference between the lower (≥400 cm) and the upper part of the section (0.18–0.38 and 0.10–0.17 wt.%, respectively).

Chemical as well as stable carbon and oxygen isotope compositions of bulk sediment carbonate are listed in Table 4. The oxygen isotope compositions are negatively correlated with the Sr content ($R^2=0.85$) and show a well-expressed negative correlation ($R^2=0.92$) with the Xcalcite value (calcite/(calcite+dolomite)) (Fig. 3D). The carbon isotope compositions of the bulk carbonate are negatively correlated with the carbonate content ($R^2=0.60$), whereas the correlation with the Sr content is not that straightforward ($R^2=0.28$). The carbon isotope – Xcalcite negative correlation is significant only if the samples below 400 cm are excluded ($R^2=0.70$) (Fig. 3D). The concentrations of other mineral components and elements show either random or rather complex patterns.

 TiO_2 and P_2O_5 contents of bulk sediments are higher in the Holocene (<10 ky) parts than in the Pleistocene layers (Fig. 4A) where they display similar patterns. The Sr concentration of the bulk sediment is 160 to 180 ppm below 400 cm, then decreases to about 120 ppm up-section (Fig. 4B).

The relationship between Xcalcite values and isotopic compositions of bulk carbonate determines the composition of the dolomite component with $\delta^{13}C \sim +3\%$ and $\delta^{18}O \sim -2\%$. Using the Xcalcite and Xdolomite values measured by



Fig. 3

52 50

0

5000

10000

age (yr cal BP)

15000

Calcite content of the carbonate fraction (A) and dolomite (B) and SiO_2 (C) contents of the bulk sediment of the Körösladány section as a function of age. (D) Stable carbon (squares) and oxygen (circles) isotope compositions of bulk carbonate vs. calcite content of the total carbonate fraction. Regression calculation was conducted for the carbon isotope compositions excluding the >16.5 ky samples due to the very different sedimentological characteristics of the section part (empty squares) (see text and Fig. 3A–C, grey shaded bars)

20000

-10

0.3

0.4

0.5

XRD analyses and the estimated dolomite compositions, δ^{13} C and δ^{18} O values can be calculated for the calcite component using the simple mass balance equation

 δ bulk=Xcalcite $\cdot\delta$ calcite+Xdolomite $\cdot\delta$ dolomite

from which δ calcite=(δ bulk-Xdolomite· δ dolomite)/Xcalcite, where δ means δ ¹³C and δ ¹⁸O values.

The calculated δ^{18} O values show a random fluctuation throughout the entire section (with an average of -9.1 ± 0.5‰), whereas the δ^{13} C values display a distinct change from about -7‰ (below 375 cm) to -13.5 ± 1.4‰ above this level (Fig. 4C–D). Interestingly, the calculated calcite δ^{13} C and δ^{18} O values are positively correlated with an R² value of 0.75 in the upper part of the section.

The carbon isotope compositions of *Unio crassus* shells fluctuate between –14.0 and –10.5‰, whereas their oxygen isotope values show a distinct pattern with <–10‰ below 150 cm and a sudden increase to –8.3‰ at 150 cm (Table 5, Fig. 4C–D). The δ^{13} C and δ^{18} O values of *Sphaerium rivicola* shells range from –11.4 to –94‰ and from –11.3 to –9.0‰, respectively (Table 5, Fig. 4C–D). Although the shells of the two bivalve species were collected from the same sediment samples

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0.7

0.8

0.9

0.6

X(calcite)

Table 4 Stable ci from Ba	arbon and o to Zr)	xygen isotop	ve compositi	ions (in % r	elative to V-	PDB) and ch	nemical com	positions (in	ı wt.% from	. SiO ₂ to P ₂ C	l _{5,} in ppm
cm	δ ¹⁸ Ο	δ ¹³ C	SiO ₂	Al ₂ O ₃	Fe ₂ O ₃	MgO	CaO	K20	TiO ₂	MnO	P ₂ O ₅
25	-5.3	-3.9	59.1	12.3	4.4	2.4	4.0	2.6	0.69	0.13	0.30
50	-5.0	4.3	62.1	12.3	4.4	2.3	4.1	2.3	0.69	0.12	0.27
75	-5.4	-5.7	60.3	12.0	4.2	2.4	5.5	2.6	0.65	0.12	0.30
100	-5.7	-6.2	58.3	12.4	4.3	2.5	6.3	2.8	0.62	0.11	0.30
125	-6.0	-6.2	59.9	12.2	4.3	2.4	6.1	2.5	0.66	0.17	0.32
150	-5.9	-5.9	59.0	12.0	4.1	2.6	6.7	2.7	0.65	0.14	0.29
175	-5.9	-5.8	57.3	12.2	4.2	2.5	6.0	2.3	0.62	0.17	0.25
200	-5.6	-5.2	57.5	12.0	4.2	2.5	6.2	2.3	0.61	0.12	0.32
225	-6.0	-5.9	58.1	12.4	4.3	2.7	6.1	2.5	0.68	0.11	0.25
250	-6.1	-6.2	58.5	11.9	4.2	2.9	6.7	2.4	0.61	0.13	0.23
275	-6.2	-6.6	57.9	11.8	4.1	2.9	8.2	2.6	0.63	0.10	0.18
300	-6.1	-6.6	56.1	12.0	4.0	2.8	8.4	2.5	0.59	0.10	0.20
325	-6.7	-7.8	56.4	11.7	4.0	3.2	8.7	2.4	0.60	0.11	0.19
350	-6.4	-7.2	57.9	11.9	4.1	3.4	8.3	2.5	0.62	0.12	0.21
375	-6.5	-7.6	52.5	11.7	4.1	3.4	8.4	2.5	0.64	0.11	0.20
400	-7.8	-5.6	64.4	12.0	4.3	1.7	4.8	2.2	0.57	0.18	0.26
425	-8.2	-5.7	9.99	12.7	4.4	1.9	4.7	2.2	0.65	0.23	0.25
450	-7.6	-5.2	68.6	11.9	4.1	1.8	4.1	2.1	0.63	0.23	0.24
475	-7.6	-5.5	67.3	11.7	4.2	1.7	4.5	2.0	0.64	0.38	0.25
500	-8.0	7 8	57 4	12.5	45	18	66	2.4	0.61	032	0.19

Zr	337	339	314	323	321	319	315	317	318	310	295	297	295	309	299	249	291	307	369	192
Zn	91	86	84	84	82	75	80	78	86	87	78	78	71	68	70	78	71	68	70	84
Y	30	34	29	31	34	33	31	26	30	25	27	30	30	29	29	27	27	25	35	27
Λ	87	87	82	86	87	84	86	85	86	82	81	79	79	82	80	75	85	79	83	84
Sr	129	121	128	133	139	136	141	139	145	139	149	153	167	164	163	177	169	171	163	181
Sc	8.1	7.4	6.7	9.9	6.7	6.1	7.6	8.3	7.1	5.8	7.5	6.7	6.7	6.7	6.8	8.6	8.1	8.7	7.5	7.0
Rb	101	105	101	95	106	101	94	102	103	96	88	98	89	80	83	71	80	71	74	84
Pb	26		23						30	25						32	31	27		39
Ni Pb	41 26	37	43 23	40	43	41	42	37	43 30	41 25	40	40	38	36	40	42 32	43 31	45 27	41	47 39
Ga Ni Pb	17 41 26	17 37	21 43 23	13 40	13 43	16 41	15 42	37	21 43 30	41 25	40	15 40	11 38	10 36	11 40	16 42 32	10 43 31	12 45 27	21 41	11 47 39
Co Ga Ni Pb	11 17 41 26	10 17 37	9 21 43 23	10 13 40	9 13 43	9 16 41	9 15 42	9 37	10 21 43 30	10 41 25	9 40	9 15 40	8 11 38	10 10 36	9 11 40	10 16 42 32	12 10 43 31	10 12 45 27	11 21 41	10 11 47 39
Ce Co Ga Ni Pb	57 11 17 41 26	55 10 17 37	49 9 21 43 23	56 10 13 40	54 9 13 43	49 9 16 41	58 9 15 42	53 9 37	61 10 21 43 30	66 10 41 25	57 9 40	69 9 15 40	56 8 11 38	55 10 10 36	52 9 11 40	55 10 16 42 32	60 12 10 43 31	61 10 12 45 27	58 11 21 41	52 10 11 47 39
Ba Ce Co Ga Ni Pb	357 57 11 17 41 26	347 55 10 17 37	317 49 9 21 43 23	337 56 10 13 40	365 54 9 13 43	337 49 9 16 41	369 58 9 15 42	353 53 9 37	376 61 10 21 43 30	391 66 10 41 25	370 57 9 40	355 69 9 15 40	299 56 8 11 38	316 55 10 10 36	317 52 9 11 40	368 55 10 16 42 32	440 60 12 10 43 31	357 61 10 12 45 27	363 58 11 21 41	374 52 10 11 47 39

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Table 4 cont.



Fig. 4

(A) TiO₂ (with 3-point running mean, thick solid line) and P_2O_5 contents, and (B) Sr contents of bulk sediment samples and stable carbon (C) and oxygen (D) isotope compositions of *Unio crassus* and *Sphaerium rivicola* shells as well as calculated compositions of the calcite fraction of bulk sediments (see text) as a function of age. Grey shaded bars denote Pleistocene periods with elevated P_2O_5 contents

from which bulk compositions were determined, no or very poor correlations ($R^2 < 0.22$) have been observed between the different datasets.

Discussions

Verification of the age-depth model

An effective test for the reliability of the age-depth model obtained for the selected age dataset is based on comparison with well-dated records that may serve as references. Figure 5 shows a comparison between the isotopic compositions of Unio crassus shells, P2O5 contents of bulk sediment samples and, as references, the δ^{18} O record of the GRIP ice core (Johnsen et al. 2001) and the δ^{13} C data of a stalagmite from the Sofular Cave, Turkey (Fleitmann et al. 2009). The ice core record is traditionally used as a reference in paleoclimate studies, while the Sofular record represents an exceptionally well-dated stalagmite that will be used a regional reference. There are certain agreements and discrepancies in the patterns of the records. The Pleistocene-Holocene transition is visible in the rise of the shells' isotopic compositions as well as in the phosphorus contents. The sudden drop in these values around 12.3 ky would be consistent with the Younger Dryas cooling (YD in Fig. 5), whereas the elevated phosphorus concentrations and the slight elevation in the δ^{13} C (shell) values between 12.8 and 14.0 ky compared to the samples just before and after these ages would agree with the Bølling/Allerød warm period. The older periods show discrepancies from the reference records that need to be discussed further (see later). However, the

<i>Unio crassus</i> Depth (cm)	δ ¹³ C	Std.	δ ¹⁸ Ο	Std.	<i>Sphaerium ri</i> y Depth (cm)	icola δ ¹³ C	Std.	δ ¹⁸ Ο	Std.
30	-12.83	0.65	-8.44	0.63	225	-10.34	0.12	-10.61	0.52
40	-10.77	0.91	-8.75	0.60	250	-10.40	0.21	-11.29	0.01
60	-10.50	0.63	-8.24	0.22	275	-10.56	0.37	-11.20	0.88
80	-10.78	0.78	-8.49	0.35	300	-10.59	0.20	-10.47	0.38
100	-11.57	1.20	-8.77	0.51	325	-9.56	0.18	-9.03	0.31
125	-11.83	0.93	-8.27	0.86	350	-10.40	0.04	-10.33	0.35
150	-10.91	0.84	-8.33	0.10	375	-10.11	0.37	-10.97	0.13
175	-14.03	0.38	-10.87	0.23	400	-9.99	0.85	-11.12	0.74
200	-12.56	0.84	-10.26	0.23	425	-11.45	0.21	-10.83	0.32
225	-12.16	0.60	-10.80	0.10	450	-9.96	1.16	-10.56	0.09
250	-12.24	0.53	-10.54	0.28	475	-10.33	0.70	-10.12	0.52
275	-12.94	0.03	-10.75	0.35	500	-9.40	0.35	-10.83	0.44
300	-11.95	0.45	-10.64	0.21	520	-10.34	0.30	-11.07	0.49
325	-11.19	0.16	-9.73	0.37					
350	-11.98	0.42	-10.84	0.24					
375	-11.62	0.30	-10.56	0.11					
400	-11.31	0.43	-10.12	0.44					
425	-12.04	0.95	-10.61	0.52					
450	-11.94	0.13	-10.32	0.08					
475	-12.84	0.14	-10.90	0.27					
500	-12.14	0.32	-10.56	0.20					
520	-12.48	0.29	-11.19	0.41					

Table 5 Stable carbon and oxygen isotope compositions (in ‰ relative to V-PDB) of *Unio crassus* and *Sphaerium rivicola* shells. The compositions are



Fig. 5

Stable carbon and oxygen isotope compositions of *Unio crassus* shells, P_2O_5 contents of bulk sediments, stable oxygen isotope compositions of the GRIP ice core (Johnsen et al. 2001; relative to V-SMOW) and carbon isotope compositions of a stalagmite from the Sofular cave, Turkey (Fleitmann et al. 2009; (relative to V-PDB), respectively. YD: Younger Dryas, B/A: Bolling/Allerod

present comparison indicates that the age-depth model established for the Körösladány section is sufficiently precise for a more detailed discussion on paleoclimate conditions.

Sedimentary changes

Beside the temperature and humidity variations related to the climate change processes of the last 20 ky, the variations in mineralogical and chemical compositions indicate changes in depositional environment or denudation source as well. This assumption is supported by the observation that the bulk carbonate's

isotopic compositions (which may depend on the amount and origin of detrital carbonate) show no relationships with the shell data (which may depend more on climate).

Between 16.5 and 19 ky the composition of the sediment was different from that of the subsequent period, with a larger amount of SiO₂, lower amount of dolomite, and calcite domination of the bulk carbonate (Fig. 3A-C). The good relationships between the bulk carbonate isotopic compositions and the mineralogical (e.g. calcite content in the bulk carbonate, see Fig. 3D) and chemical compositions (e.g. the negative correlation between bulk carbonate δ^{18} O values and Sr contents with an \mathbb{R}^2 value of 0.85) suggest mixing of different components. These relationships show that denudation was dominated by siliceous and limestone-derived detrital material before 16.5 ky; thereafter a dolomitic terrain began to be eroded, later with some calcitic components in the sediments. The source area was most probably the Carpathian Mts., where the river originated. The good $\delta^{13}C-\delta^{18}O$ correlation in the calcite component and the constant decrease in the carbonate content along with the Xcalcite value may indicate mixing of different calcite types (detrital calcite from sedimentary rocks or from soils, authigenic calcite formed in the fluvial sediments, and secondary calcite formed by alteration of biogenic aragonite). The inferred mixing may explain the fluctuation in isotopic compositions of the calcite component; hence only the major Pleistocene-Holocene climate change may be reflected in them.

The observed change in sediment properties may be interpreted as a sign of variation in river pattern and erosion provenance. Reorganization of the fluvial system of the area took place in several steps during the Late Pleniglacial (Gábris and Nádor 2007; Kasse et al. 2010) due to climate-related variations in humidity. The chemical compositions are a consequence of different processes related to the late glacial climate changes. The bulk sediments' stable oxygen isotope compositions are positively correlated with the Sr contents and show a good correspondence with the global warming that started about 18 ky ago (see the compilation of Shakun and Carlson 2010). The Sr content change can nicely be correlated with global signals like the δ^{18} O rise (Blunier and Brook 2001) and the CO2 increase (Schmitt et al. 2012) in Antarctic ice cores, as well as records of regional significance such as the δ^{18} O dataset of the Sofular cave stalagmite (Fleitmann et al. 2009) and the July mean temperature change in the Carpathians determined by Tóth et al. (2012). This indicates that carbonate denudation is related to temperature change that could have been responsible for the intensity of physical breakdown of limestone and dolomite. On the other hand, the TiO₂ concentration in the bulk sediment shows a major increase during the Pleistocene-Holocene transition as well as during the Bølling/Allerød warm period (Fig. 4A). This pattern and the correspondence between the TiO_2 and P_2O_5 changes indicate the effect of chemical weathering, the efficiency of which depends on temperature and humidity. A more humid climate is indicated by elevated phosphorus content in the fluvial sediment and the higher Ti contents

in these periods were produced by increased chemical weathering. These observations show that the chemical composition of fluvial sediments can reflect both the long term-global temperature rise and the millennial scale temperature/ humidity changes.

Climate-related changes

The next question is whether climate-related isotopic variations can be detected in shells and bulk carbonates. As a usual reference the isotopic compositions are compared to the oxygen isotope record of the GRIP ice core (Fig. 5). The most obvious effect is related to the Pleistocene-Holocene change, which is appropriately reflected by the δ^{18} O values of Unio shells (Fig. 5). The oxygen isotope composition of the calcite component of the bulk carbonate does not show a real change from the Pleistocene to the Holocene (Fig. 4D), which can be related to the mixing of different calcite components (detrital, authigenic and secondary calcite). The rather incoherent picture obtained from the isotopic compositions of bivalve shells in the Pleistocene can be related to the complex interplay of temperature of shell formation and water composition related to ambient air temperature and humidity (see the study of Schöll-Barna et al. 2012, on the Unio shells of Lake Balaton). A rather significant positive δ^{18} O peak appears both in the Unio and the Sphaerium shells at about 15.5 ky (Fig. 4C–D). This period just precedes the Bølling/Allerød warm period and can be attributed to the Oldest Dryas cold event (Alley and Clark 1999). Since these cold events (such as the Younger Dryas event; Magyari et al. 2012a) are also characterized by rather arid conditions, the strong evaporation under dry conditions could have been responsible for the coupled positive δ^{13} C and δ^{18} O shifts in the bivalve shells at about 15.5 ky. The assumed aridity is also supported by the low phosphorus content indicating decreased biogenic activity. Although this period was cold, temperature decrease would not serve as an explanation for the elevated $\delta^{18}O$ values, as lower ambient temperatures would cause lower δ^{18} O values in the precipitation water. Since the δ^{18} O–T gradient exceeds the effect of temperature decrease on the carbonate-water oxygen isotope fractionation (see Schöll-Barna et al. 2012), the net result of temperature decrease would be lower δ^{18} O values in the carbonate.

The slightly elevated δ^{13} C (shell) values and the increased phosphorus contents around 17–18 ky indicate another warm and humid period that is not represented in the ice core and stalagmite data described above. This period roughly corresponds to the Ságvár-Lascaux interstadial defined by Sümegi and Krolopp (2002). Since it may have been a local climate phenomenon, comparison with local paleoclimate data compilations could provide an insight into the significance of this event. Paleotemperature reconstructions spanning the entire period from the Last Glacial Maximum to the Holocene at sufficiently good age resolution are rather scarce (Sümegi and Krolopp 2002; Gábris and Nádor 2007). The bulk sediment P_2O_5 concentrations are compared with the paleotemperature data of Gábris and Nádor (2007) in Fig. 6, although we note that some of the data compiled by Gábris and Nádor (2007) are based on uncertain age dating. Again, the most prominent feature is the good correspondence between these records during the Pleistocene-Holocene transition and the Ságvár-Lascaux interstadial. Interestingly, the Ságvár-Lascaux interstadial seems to be more significant than the Bølling/Allerød period, both in the paleotemperature curve of Gábris and Nádor (2007) and in the malacothermometer data obtained for the southern part of the Great Hungarian Plain (Sümegi 2005), whereas the phosphorus concentrations are rather high in the younger period related to increased humidity compared to the Ságvár-Lascaux interstadial.

An attempt was made to compare the C and O isotope data of bivalve shells with the paleotemperature records of Sümegi (2005) and Gábris and Nádor (2007), but the comparison yielded an incomprehensible picture due to competing fractionation processes. The δ^{13} C value of shell aragonite depends mainly on the carbon isotope composition of dissolved inorganic carbon (DIC) in the water and the amount of metabolic carbon incorporated by the shell (Dettman et al. 1999; Veinott and Cornett 1996; Lorrain et al. 2004; Gillikin et al. 2009). In a fluvial environment Zhou et al. (2010) detected C isotope changes in the DIC related to biogenic activity, with DIC enriched in ¹³C in highly eutrophic water bodies due



Fig. 6

Comparison of P_2O_5 contents of bulk sediments of the Körösladány section and paleotemperature data of Sümegi (2005) and Gábris and Nádor (2007)

to the ¹²C-removal by the organic matter. The δ^{18} O value of shell aragonite depends on the ambient temperature and the oxygen isotope composition of the water (Grossman and Ku 1986; Dettman et al. 1999). Water composition can be calculated from the measured shell δ^{18} O values using the paleotemperature data of Gábris and Nádor (2007) and the shell-water fractionation equation of Dettman et al. (1999). If the temperature variations are preserved in the δ^{18} O (shell) values, realistic water composition should be obtained. In theory, at constant water composition the changes in δ^{18} O value of shells would be determined entirely by temperature variations. On the other hand, calculation of water composition would eliminate the temperature effect, and fluctuations in the water value would be related to paleohydrological changes. The calculated water composition ranged from -12.7 to -8.0% (Fig. 7), in good agreement with published values (from -14 to -8‰; Varsányi et al. 2011). For precipitation water there is a well-known relationship between the water's oxygen isotope composition and the ambient temperature that depends on the local climate conditions. The $\delta^{18}O/\Delta T$ coefficient is 0.37 for the Great Plain (Deák 1995 cited in Fórizs 2003; Vodila et al. 2011), whereas the slope of the relationship is higher for the precipitation in the Carpathians (Kern et al. 2009) from where most of the Tisza waters originate. Plotting the calculated water composition against paleotemperature revealed a relatively elevated $\Delta^{18}O/\Delta T$ coefficient above 16 °C and quasi constant $\delta^{18}O$ (water) values below 16 °C (Fig. 7), raising the possibility of a changing $\Delta^{18}O-\Delta T$



Fig. 7

Calculated water compositions (relative to V-SMOW) (from the data of Unio and Sphaerium shells, respectively, see text) as a function of paleotemperature data of Gábris and Nádor (2007)

equation. The observed changes indicate that a strong $\Delta^{18}O-\Delta T$ relationship existed for the warm periods (the Holocene, the Bølling/Allerød and the Ságvár-Lascaux interstadials), whereas the $\delta^{18}O$ (water) values showed no significant change with temperature during the cold periods. This can be explained by assuming a buffer of massive snow cover in the Carpathians during the cold periods, whose summer melting produced a continuous input of ¹⁸O-depleted precipitation, providing a buffer for the O isotope composition of the Tisza river during the cold periods. On the other hand, at periods above 16 °C mean summer temperature, the calculated water composition shows a strong dependence on ambient temperature, similar to meteoric waters in the Carpathian Mountains. Interestingly, the data belonging to the Ságvár-Lascaux interstadial also indicate an elevated $\Delta^{18}O-\Delta T$ relationship, which means a significant reduction in the summer snow cover in the Carpathians.

Conclusions

A 5.2 m-thick fluvial sequence in southeastern Hungary was studied by means of mineralogical, geochemical and stable C and O isotope analyses of bulk sediments as well as stable C and O isotope analyses of Unio crassus and Sphaerium rivicola shells. On the basis of ¹⁴C age dating, the section covers the period of about 5-20 ky BP. Effects of sedimentary changes could be detected in the chemical and isotopic compositions of fluvial sediments. The chemical composition of the bulk sediments indicated changes in denudation style from physical erosion of carbonate rocks (affecting bulk carbonate isotope compositions and bulk sediment Sr concentrations) due to a long-term temperature rise during the latest glacial, to chemical weathering of silicate rocks (determining the Ti concentrations) attributed to humidity increase indicated by an elevated phosphorus content of bulk sediment samples. Comparing the data with local paleotemperature reconstructions, it was concluded that the C and O isotope compositions of Unio and Sphaerium shells reflect climate conditions. Beside the globally important Bølling/Allerød and Younger Dryas periods, the locally significant Ságvár-Lascaux interstadial appears significant in the isotopic and chemical records. Calculated $\Delta^{18}O-\Delta T$ relationships indicate that during the cold periods the snow cover in the Carpathian Mts. served as a low- δ^{18} O buffer for the river water, whereas warmer periods (including the Ságvár-Lascaux interstadial) were associated with strong reduction of summer snow cover.

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