

## Synthesis of Phosphonates in a Continuous Flow Manner

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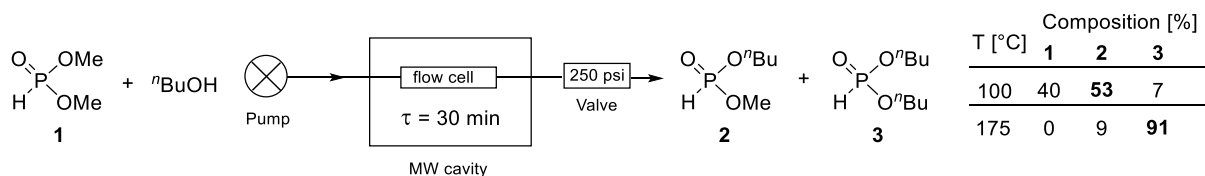
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**Abstract** The synthesis of dialkyl *H*-phosphonates and  $\alpha$ -aminophosphonates was studied in a continuous flow microwave reactor. Depending on the conditions, the alcoholysis of dialkyl *H*-phosphonates could be fine-tuned towards the mixed and the fully transesterified products. The continuous flow synthesis of  $\alpha$ -aryl- $\alpha$ -aminophosphonates was elaborated utilizing the aza-Pudovik reaction of imines and dialkyl *H*-phosphonates, as well as the by the Kabachnik-Fields condensation of primary amines, benzaldehyde and  $>P(O)H$  reagents.

**Keywords** dialkyl *H*-phosphonates;  $\alpha$ -aminophosphonates; alcoholysis; Kabachnik-Fields reaction; Pudovik reaction; continuous flow reactor

Microwave (MW)-assistance is a useful technique, however, the size of the reactor is rather limited.<sup>1</sup> MW-assisted reactions on a bigger scale may be carried out in a continuous flow MW equipment.<sup>2</sup> In this work, the synthesis of phosphonates was investigated in a self-developed continuous flow MW system based on a CEM Discover MW reactor equipped with a commercially available CEM continuous flow cell.

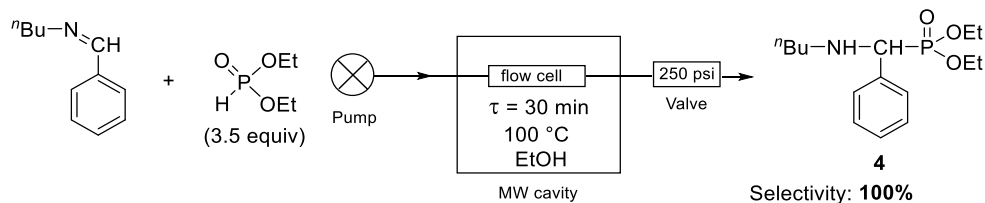
The continuous flow alcoholysis of dimethyl *H*-phosphonate (**1**) with *n*-butanol (Scheme 1) was studied in the single pump system based on our experiences in a batch MW reactor.<sup>3</sup> At a residence time of 30 min and 100°C, the *n*-butyl methyl *H*-phosphonate (**2**) was obtained as the main component (53%), while at 175 °C, the di(*n*-butyl) *H*-phosphonate (**3**) predominated (91%). Further increase of the temperature did not change the composition.



**Scheme 1** Continuous flow alcoholysis of dimethyl *H*-phosphonate with *n*-butanol.

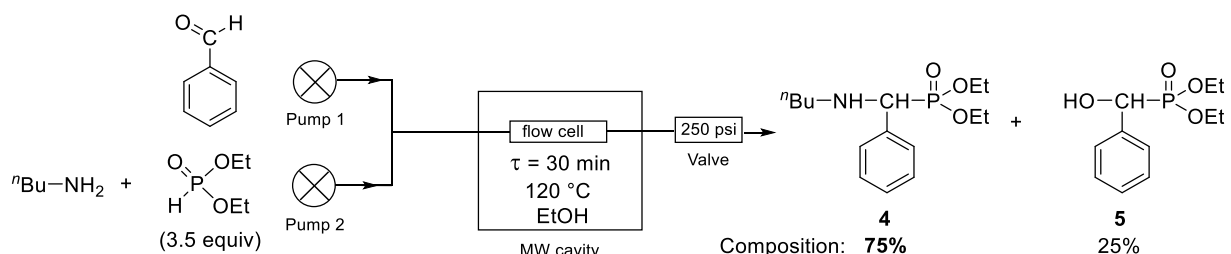
An efficient method has been developed for the synthesis of  $\alpha$ -aryl- $\alpha$ -aminophosphonates (**4**) by the catalyst- and solvent-free MW-assisted aza-Pudovik reaction in a batch MW reactor.<sup>4</sup> To make the procedure flow compatible, a preliminary experiment was carried out for the addition of diethyl phosphite

to *N*-benzylidene(*n*-butyl)amine in ethanol as the solvent (Scheme 2). The diethyl ((*n*-butylamino)(phenyl)methyl)phosphonate (**4**) was formed selectively.



**Scheme 2** Continuous flow synthesis of diethyl ((*n*-butylamino)(phenyl)methyl)phosphonate by aza-Pudovik reaction.

As an extension, after changing for the dual pump system, the same  $\alpha$ -aminophosphonate (**4**) was prepared by the three-component Kabachnik-Fields reaction of *n*-butylamine, diethyl phosphite and benzaldehyde in ethanol without any catalyst (Scheme 3). Although, the aldehyde and the phosphite were pumped from separated vessels, beside the desired product (**4**) (75%), 25% of the  $\alpha$ -hydroxyphosphonate (**5**) was also formed.



**Scheme 3** Continuous flow reaction of *n*-butylamine, diethyl phosphite and benzaldehyde.

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## REFERENCES

1. Bálint, E.; Keglevich, G. The Spread of the Application of the Microwave Technique in Organic Synthesis, In: Keglevich, G. (ed.) *Milestones in Microwave Chemistry*, Springer: Switzerland, 2016; pp 1-10.
2. Kappe, C. O.; Stadler, A.; Dallinger, D. *Microwaves in organic and medicinal chemistry*, Wiley: Weinheim, 2012.
3. Bálint, E.; Tajti, Á.; Drahos, L.; Ilia, G.; Keglevich, G. *Curr. Org. Chem.*, **2013**, 17, 555-562.
4. Bálint, E.; Tajti, Á.; Ádám, A.; Csontos, I.; Karaghiosoff, K.; Czugler, M.; Ábrányi-Balogh, P.; Keglevich, G. *Beilstein J. Org. Chem.*, **2017**, 13, 76-86.

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